



Government of Maharashtra  
**ISMAIL YUSUF COLLEGE OF ARTS, SCIENCE AND COMMERCE**  
**NAAC 'A' Grade**  
**Jogeshwari (East), Mumbai 400 060**

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Department of Chemistry

M.Sc. Part I

Semester I January 2022

| Sr. No | Name of Subject                             | Date and Time                      | Name of Examiner        |
|--------|---|------------------------------------|-------------------------|
| 1      | Physical Chemistry: Course Code: PSCH 101   | 08.01.2022<br>(09:00am to 05:00pm) | Shri. Yatendra Yadav    |
| 2      | Inorganic Chemistry: Course Code: PSCH 102  | 10.01.2022<br>(09:00am to 05:00pm) | Miss. Saira Siddique    |
| 3      | Organic Chemistry: Course Code: PSCH 103    | 11.01.2022<br>(09:00am to 05:00pm) | Miss. Ruqaiya Ansari    |
| 4      | Analytical Chemistry: Course Code: PSCH 104 | 12.01.2022<br>(09:00am to 05:00pm) | Shri Pramod Vishwakarma |

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Department of Chemistry

  
**Head of Department**  
**CHEMISTRY**

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**Ismail Yusuf College Jogeshwari(E), Mumbai**

Department of Chemistry  
M. Sc. I Sem-I (Internal Presentation Topics) October-2021

| Sr. No. | Name of Students             | Paper No. | Topic For presentation   |
|---------|------------------------------|-----------|--|
| 1.      | Khan Iqra Azizullah          | Paper-I   | Maxwell equations, Maxwell thermodynamic Relations; its significance and applications to ideal gases.                    |
|         |                              | Paper-II  | Preparative methods: Chemical methods, Solvothermal,   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: rate and equilibrium constants                                     |
|         |                              | Paper-IV  | Laser as a source of radiation, Fibre optics   |
| 2.      | Khan Khushnuma Abdul Moid    | Paper-I   | Third law of Thermodynamics, Entropy change for a phase transition, absolute entropies                                   |
|         |                              | Paper-II  | Preparative methods:, Solvothermal, Combustion synthesis,  |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: reaction coordinate diagram, transition state (activated complex), |
|         |                              | Paper-IV  | Introduction of Fourier Transform.   |
| 3.      | Khan Mehjabeen Mohammad      | Paper-I   | Joule Thomson experiment, Joule Thomson coefficient  |
|         |                              | Paper-II  | Preparative methods: Microwave, Co-precipitation,  |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: nature of activated complex, Hammond postulate,                    |
|         |                              | Paper-IV  | Derivation of Beer- Lambert's Law and its limitations  |
| 4.      | Khan Tabassum Ejaz           | Paper-I   | Joule Thomson coefficient in terms of van der Waals constants  |
|         |                              | Paper-II  | Preparative methods: Langmuir Blodgett (L-B) method,   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: Reactivity vs selectivity,   |
|         |                              | Paper-IV  | Factors affecting molecular absorption, types of transitions   |
| 5.      | Khan Zainab Parvin Noor      | Paper-I   | Standard molar entropies and their dependence on molecular mass and molecular structure                                  |
|         |                              | Paper-II  | Biological methods: Synthesis using microorganisms   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: Curtin-Hammett Principle,  |
|         |                              | Paper-IV  | factors affecting molecular absorption: pH, temperature,   |
| 6.      | Kumbhar Vaishnavi Shashikant | Paper-I   | Classical Mechanics, failure of classical mechanics  |
|         |                              | Paper-II  | Applications in the field of semiconductors,   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: Microscopic reversibility,   |



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|     |                                       |                  |  |
|-----|---------------------------------------|------------------|--|
|     |                                       | <b>Paper-IV</b>  | Factors affecting molecular absorption: solvent and effect of substituents.  |
| 7.  | Mahadeshwar<br>Diksha Dilip           | <b>Paper-I</b>   | Need for Quantum Mechanics.  |
|     |                                       | <b>Paper-II</b>  | Applications in the field of solar cells   |
|     |                                       | <b>Paper-III</b> | Thermodynamic and kinetic requirements of a reaction: Kinetic vs thermodynamic control of organic reactions.                                   |
|     |                                       | <b>Paper-IV</b>  | Dual spectrometry – Introduction, Principle, Instrumentation and Applications  |
| 8.  | Mansoori Safa<br>Banu Chand<br>Farooq | <b>Paper-I</b>   | Particle waves and Schrödinger wave equation   |
|     |                                       | <b>Paper-II</b>  | Characterisation of Coordination compounds: Formation, thermal studies, Conductivity measurement   |
|     |                                       | <b>Paper-III</b> | Ester hydrolysis: Classification   |
|     |                                       | <b>Paper-IV</b>  | Applications of Ultraviolet and Visible spectroscopy: 1) On charge transfer absorption   |
| 9.  | Mansuri Md<br>Jahangir<br>Nizamuddin  | <b>Paper-I</b>   | wave functions, properties of wave functions   |
|     |                                       | <b>Paper-II</b>  | Characterisation of Coordination compounds: electronic spectral and magnetic measurements  |
|     |                                       | <b>Paper-III</b> | Determining mechanism of a reaction: Product analysis,   |
|     |                                       | <b>Paper-IV</b>  | Applications of Ultraviolet and Visible spectroscopy: Simultaneous spectroscopy  |
| 10. | Mishra Nishant<br>Deoprasaksh         | <b>Paper-I</b>   | Normalization of wave functions,   |
|     |                                       | <b>Paper-II</b>  | Characterisation of Coordination compounds: IR   |
|     |                                       | <b>Paper-III</b> | Determining mechanism of a reaction: kinetic studies, use of isotopes (Kinetic isotope effect – primary and secondary kinetic isotope effect). |
|     |                                       | <b>Paper-IV</b>  | Applications of Ultraviolet and Visible spectroscopy: Derivative Spectroscopy  |
| 11. | Narkar Ankita<br>Sadashiv             | <b>Paper-I</b>   | Orthogonality of wave functions,   |
|     |                                       | <b>Paper-II</b>  | Characterisation of Coordination compounds: NMR  |
|     |                                       | <b>Paper-III</b> | Determining mechanism of a reaction: Detection and trapping of intermediates,  |
|     |                                       | <b>Paper-IV</b>  | Oxidation number, rules for assigning oxidation number,  |
| 12. | Neman Kunal<br>Eknath                 | <b>Paper-I</b>   | Operators and their algebra  |
|     |                                       | <b>Paper-II</b>  | Preparative methods: Chemical methods  |
|     |                                       | <b>Paper-III</b> | Characterisation of Coordination compounds: ESR  |
|     |                                       | <b>Paper-IV</b>  | Redox reaction in term of oxidation number, oxidizing and reducing agents,   |
| 13. | Pandhare Sudhir<br>Bhanudas           | <b>Paper-I</b>   | Linear and Hermitian operators   |
|     |                                       | <b>Paper-II</b>  | Microwave synthesis (discussion on principles, examples, merits and demerits are expected)   |
|     |                                       | <b>Paper-III</b> | Determining mechanism of a reaction: crossover experiments and stereochemical evidence,  |
|     |                                       | <b>Paper-IV</b>  | Equivalent weight of oxidizing and reducing agents, stoichiometry of redox titration   |

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|-----|----------------------------------|-----------|---|
| 14. | Shaikh Abu Bakar<br>Bakar Husain | Paper-I   | Operators for the dynamic variables of a system such as, position, linear momentum, angular momentum                          |
|     |                                  | Paper-II  | Methods of preparation for inorganic solids: Ceramic method, precursor method, sol-gel method (applications in Biosensors).   |
|     |                                  | Paper-III | Determining mechanism of a reaction: stereochemical evidence.   |
|     |                                  | Paper-IV  | Normality of a solution of a oxidizing / reducing agent and its relationship with molarity.                                   |
| 15. | Shaikh Arbaaz<br>Siraj           | Paper-I   | Operators for the dynamic variables of a system such as, total energy, eigen functions, eigen values and eigen value equation |
|     |                                  | Paper-II  | Layer structure [cadmium chloride and iodide ( $\text{CdCl}_2$ , $\text{CdI}_2$ )].   |
|     |                                  | Paper-III | Acids and Bases: Factors affecting acidity and basicity: Electronegativity  |
|     |                                  | Paper-IV  | Calculations of pH of acids, bases, acidic and basic buffers.   |
| 16. | Shaikh Misba<br>Sakhavat         | Paper-I   | Schrödinger wave equation as the eigen value equation of the Hamiltonian operator   |
|     |                                  | Paper-II  | Electronic structure of solids and band theory  |
|     |                                  | Paper-III | Acids and Bases: Factors affecting acidity and basicity: inductive effect   |
|     |                                  | Paper-IV  | Calculations of pH of acids, bases, acidic and basic buffers.   |
| 17. | Shaikh Mohammed<br>Ammar Ilyas   | Paper-I   | Average value and the expectation value of a dynamic variable of the system   |
|     |                                  | Paper-II  | Fermi level, K Space and Brillouin Zones.   |
|     |                                  | Paper-III | Acids and Bases: Factors affecting acidity and basicity: resonance, bond strength, electrostatic effects.                     |
|     |                                  | Paper-IV  | Solubility and solubility equilibria, effect of presence of common ion.   |
| 18. | Shaikh Tubba<br>Ansar            | Paper-I   | Postulates of Quantum Mechanics   |
|     |                                  | Paper-II  | Structures of Compounds of the type: AB [nickel arsenide ( $\text{NiAs}$ )], $\text{AB}_2$ [fluorite ( $\text{CaF}_2$ )]      |
|     |                                  | Paper-III | Acids and Bases: Factors affecting acidity and basicity: hybridization, aromaticity and solvation.                            |
|     |                                  | Paper-IV  | Stoichiometry of chemical reactions, concept of kg mol, limiting reactant, theoretical and practical yield.                   |
| 19. | Haikh Zainabfiya<br>Imtiaz       | Paper-I   | Schrodinger's Time independent wave equation  |
|     |                                  | Paper-II  | Structures of Compounds of the type: AB [nickel arsenide ( $\text{NiAs}$ )].)   |
|     |                                  | Paper-III | Comparative study of acidity and basicity of organic compounds on the basis of $\text{pK}_a$ values.                          |
|     |                                  | Paper-IV  | Calculations of ppm, ppb and dilution of the solutions, concept of mmol   |



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|-----|------------------------------|-----------|---|
| 20. | Singh Shrasti<br>Manoj Kumar | Paper-I   | Schrodinger's time dependent wave equation.   |
|     |                              | Paper-II  | Structures of Compounds of the type: AB <sub>2</sub> [fluorite (CaF <sub>2</sub> )  |
|     |                              | Paper-III | Leveling effect and non-aqueous solvents.   |
|     |                              | Paper-IV  | Nomenclature of all eight mechanisms of acid and base catalyzed hydrolysis.   |
| 21. | Vhadade Amar<br>Shripati     | Paper-I   | Application of quantum mechanics to the following systems: a) Free particle, wave function and energy of a free particle. |
|     |                              | Paper-II  | Structures of Compounds of the type: AB <sub>2</sub> rutile (TiO <sub>2</sub> ) structure                                 |
|     |                              | Paper-III | Acid and base catalysis – general and specific catalysis with examples.   |
|     |                              | Paper-IV  | Concentration of a solution based on volume and mass units.   |

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Department of Chemistry  
**M. Sc. I Sem-I (Internal Marks) October-2021**

| Sr. No. | Name of Students              | Paper-I | Paper-II | Paper-III | Paper-IV |
|---------|-------------------------------|---------|----------|-----------|----------|
| 1.      | Khan Iqra Azizullah           | 29      | 26       | 28        | 30       |
| 2.      | Khan Khushnuma Abdul          | 32      | 26       | 31        | 30       |
| 3.      | Khan Mehjabeen Mohammad Shafi | 30      | 24       | 22        | 31       |
| 4.      | Khan Tabassum Ejaz            | 31      | 25       | 23        | 29       |
| 5.      | Khan Zainab Parvin Noor       | 33      | 26       | 26        | 30       |
| 6.      | Kumbhar Vaishnavi Shashikant  | 32      | 27       | 27        | 31       |
| 7.      | Mahadeshwar Diksha Dilip      | 27      | 26       | 24        | 32       |
| 8.      | Mansoori Safa Banu Chand      | 32      | 26       | 26        | 30       |
| 9.      | Ansuri Md Jahangir Nizamuddin | 30      | 26       | 25        | 29       |
| 10.     | Mishra Nishant Deopraakash    | 35      | 30       | 29        | 32       |
| 11.     | Narkar Ankita Sadashiv        | 27      | 24       | 27        | 28       |
| 12.     | Neman Kunal Eknath            | 32      | 23       | 25        | 30       |
| 13.     | Pandhare Sudhir Bhanudas      | 30      | 24       | 23        | 29       |
| 14.     | Shaikh Abu Bakar Bakar Husain | 31      | 25       | 22        | 30       |
| 15.     | Shaikh Arbaaz Siraj           | 27      | 22       | 27        | 30       |
| 16.     | Shaikh Misba Sakhat           | 30      | 26       | 28        | 31       |
| 17.     | Shaikh Mohammed Ammar Ilyas   | 31      | 26       | 25        | 31       |
| 18.     | Shaikh Tubba Ansar            | 30      | 25       | 24        | 30       |
| 19.     | Shaikh Zainabfiya Imtiaz      | 35      | 32       | 27        | 31       |
| 20.     | Singh Shrasti Manoj Kumar     | 30      | 30       | 30        | 32       |
| 21.     | Vhadade Amar Shripati         | 26      | 26       | 23        | 30       |

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Department of Chemistry

M.Sc. Part II

Semester III December 2021

Physical Chemistry

| Sr. No | Subject Code | Name of Subject   | Date and Time                      | Name of Examiner            |
|--------|--------------|---|------------------------------------|-----------------------------|
| 1      | PSCHP301     | Chemistry: Polymer, Surface & Photochemistry              | 10.12.2021<br>(09:00am to 12:00pm) | Shri. Yatendra Yadav        |
| 2      | PSCHP302     | Nano chemistry, statistical mechanics & Nuclear chemistry | 10.12.2021<br>(01:00pm to 5:00pm)  | Shri. Yatendra Yadav        |
| 3      | PSCHP303     | Atomic and Molecular: Structure and Spectroscopy          | 11.12.2021<br>(09:00am to 12:00pm) | Shri. Rajeev Verma          |
| 4      | PSCHP304     | Advanced Instrumental Techniques                          | 11.12.2021<br>(01:00pm to 5:00pm)  | Shri. Pramod<br>Vishwakarma |

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Department of Chemistry

M.Sc. Part II

Semester III December 2021

Inorganic Chemistry

| Sr. No | Subject Code | Name of Subject                          | Date and Time                      | Name of Examiner     |
|--------|--------------|--|------------------------------------|----------------------|
| 1      | PSCHP301     | Chemistry of Inorganic Solids            | 10.12.2021<br>(09:00am to 12:00pm) | Miss. Ruqaiya Ansari |
| 2      | PSCHP402     | Bioinorganic and Coordination Chemistry. | 10.12.2021<br>(01:00pm to 5:00pm)  | Miss. Saira Siddique |
| 3      | PSCHP403     | Spectral Methods in Inorganic Chemistry  | 11.12.2021<br>(09:00am to 12:00pm) | Miss. Ruqaiya Ansari |
| 4      | PSCHP404     | Inorganic Materials                      | 11.12.2021<br>(01:00pm to 5:00pm)  | Miss. Saira Siddique |

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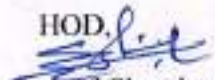
Department of Chemistry

M.Sc. Part II

Semester III December 2021

Organic Chemistry

| Sr. No | Subject Code | Name of Subject                            | Date and Time                      | Name of Examiner        |
|--------|--------------|--|------------------------------------|-------------------------|
| 1      | PSCHO301     | Theoretical organic chemistry-I            | 10.12.2021<br>(09:00am to 12:00pm) | Dr. Dattatraya Bhangare |
| 2      | PSCHO302     | Synthetic organic chemistry-I              | 10.12.2021<br>(01:00pm to 5:00pm)  | Shri. Rajeev Verma      |
| 3      | PSCHO303     | Natural products and Spectroscopy          | 11.12.2021<br>(09:00am to 12:00pm) | Shri. Rajeev Verma      |
| 4      | PSCHO304     | Medicinal , Biogenesis and green chemistry | 11.12.2021<br>(01:00pm to 5:00pm)  | Dr. Dattatraya Bhangare |

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**M.Sc-II Sem-III Topic For Presentation October-2021**

**Organic Chemistry**

| Sr. No. | Name of Students               | Paper No. | Topic For presentation  |
|---------|--------------------------------|-----------|---|
| 1.      | Harwani Saloni Sanil           | Paper-I   | Organic reactive intermediates, methods of generation, structure, stability and important reactions involving carbocations. |
|         |                                | Paper-II  | Mukaiyama esterification, Mitsunobu reaction  |
|         |                                | Paper-III | Introduction to naturally occurring sugars: Deoxysugars.  |
|         |                                | Paper-IV  | Primary and secondary metabolites and the building blocks   |
| 2.      | Khan Nikhat Parveen Ahsanullah | Paper-I   | Organic reactive intermediates, methods of generation, structure, stability and important reactions involving nitrenes.     |
|         |                                | Paper-II  | Darzen's Glycidic Ester synthesis, Ritter reaction  |
|         |                                | Paper-III | Introduction to naturally occurring sugars: Aminosugars.  |
|         |                                | Paper-IV  | General pathway of amino acid biosynthesis.   |
| 3.      | Khan Umar Farooq Khan          | Paper-I   | Organic reactive intermediates, methods of generation, structure, stability and important reactions involving carbenes.     |
|         |                                | Paper-II  | Yamaguchi esterification, Peterson olefination  |
|         |                                | Paper-III | Acetate pathway: Biosynthesis of malonylCoA   |
|         |                                | Paper-IV  | Derivation of Beer- Lambert's Law and its limitations   |
| 4.      | Sidique Mohamad Ahmad Abdul    | Paper-I   | Organic reactive intermediates, methods of generation, structure, stability and important reactions involving arynes.       |
|         |                                | Paper-II  | Domino reactions: Characteristics   |
|         |                                | Paper-III | Introduction to naturally occurring sugars: branched sugars.  |
|         |                                | Paper-IV  | Biosynthesis of saturated fatty acids.  |
| 5.      | Tiware Sarvagya Kumar Arvind   | Paper-I   | Organic reactive intermediates, methods of generation, structure, stability and important reactions involving ketenes.      |
|         |                                | Paper-II  | Nazarov cyclization   |
|         |                                | Paper-III | Structure elucidation of lactose and Dglucosamine   |
|         |                                | Paper-IV  | Biosynthesis of prostaglandins from arachidonic acid.   |
| 6.      | Yadav Anish Anil               | Paper-I   | Neighbouring group participation: Mechanism and effects of anchimeric assistance  |
|         |                                | Paper-II  | Multicomponent reactions: Strecker Synthesis, Ugi 4CC,  |
|         |                                | Paper-III | General structural features, occurrence, biological importance and applications of: carotenoids                             |



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|----|---------------------|-----------|--|
| 7. | Yadav Dinesh Suresh | Paper-IV  | Shikimic Acid pathway: Biosynthesis of shikimic acid   |
|    |                     | Paper-I   | NGP by unshared/ lone pair electrons, $\pi$ -electrons, aromatic rings                           |
|    |                     | Paper-II  | Multicomponent reactions: Biginelli synthesis, Hantzsch synthesis                                |
|    |                     | Paper-III | General structural features, occurrence, biological importance and applications of: anthocyanins |
|    |                     | Paper-IV  | Shikimic Acid pathway: Biosynthesis of aromatic amino acids, cinnamic acid and its derivatives   |

**Inorganic Chemistry**

|   |                        |           |   |
|---|------------------------|-----------|---|
| 1 | Beig Parvej Khwaja     | Paper-I   | Structures of AB type compounds (PbO and CuO)   |
|   |                        | Paper-II  | Coordination geometry of the metal ion and functions.   |
|   |                        | Paper-III | X-Ray Diffraction: Bragg Condition; Miller Indices  |
|   |                        | Paper-IV  | Handling of Hazardous Materials   |
| 2 | Kadam Pratik Prakash   | Paper-I   | Structures of AB <sub>2</sub> type ( $\beta$ -cristobalite, CaC <sub>2</sub> and Cs <sub>2</sub> O)                                 |
|   |                        | Paper-II  | Coordination geometry of the metal ion and functions.   |
|   |                        | Paper-III | Miller Indices; Laue Method   |
|   |                        | Paper-IV  | Toxic Materials Various types of toxins and their effects on humans   |
| 3 | Mane Swapnali Yuvraj   | Paper-I   | Structures of A <sub>2</sub> B <sub>3</sub> type (Cr <sub>2</sub> O <sub>3</sub> and Bi <sub>2</sub> O <sub>3</sub> )               |
|   |                        | Paper-II  | Coordination geometry of the metal ion and functions.   |
|   |                        | Paper-III | Zn in biological systems: Carbonic anhydrase.   |
|   |                        | Paper-IV  | Toxic Materials Various types of toxins and their effects on humans   |
| 4 | Qureshi Heena Jafar    | Paper-I   | Structures of B <sub>3</sub> (ReO <sub>3</sub> , Li <sub>3</sub> N)   |
|   |                        | Paper-II  | Zn in biological systems: protolytic enzymes.   |
|   |                        | Paper-III | Bragg Method; Debye Scherrer Method of X-Ray Structural Analysis of Crystals.   |
|   |                        | Paper-IV  | Explosives and Inflammable Materials  |
| 5 | Singh Ashwini Anil     | Paper-I   | Structures of ABO <sub>3</sub> type, relation between ReO <sub>3</sub> and perovskite BaTiO <sub>3</sub> and its polymorphic forms. |
|   |                        | Paper-II  | Zn in biological systems: carboxy peptidase, Zinc finger.   |
|   |                        | Paper-III | <b>Electron Diffraction:</b> Scattering of electron   |
|   |                        | Paper-IV  | Explosives and Inflammable Materials  |
| 6 | Khan Mohd Shaban Abdul | Paper-I   | Structures of ABO <sub>3</sub> type, relation between ReO <sub>3</sub> and perovskite BaTiO <sub>3</sub> and its Oxide bronzes.     |
|   |                        | Paper-II  | Role of metal ions in biological electron transfer processes: iron sulphur proteins   |
|   |                        | Paper-III | <b>Electron Diffraction:</b> Scattering Intensity versus Scattering Angle   |
|   |                        | Paper-IV  | Types of fire extinguishers(chemical reaction)  |
| 7 | Manihar Rafiullah S.   | Paper-I   | Structures of ABO <sub>3</sub> type, relation between ReO <sub>3</sub> and perovskite BaTiO <sub>3</sub> and its ilmenite structure |

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|  | Paper-II  | Role of metal ions in biological electron transfer processes; iron sulphur proteins |
|  | Paper-III | <b>Electron Diffraction:</b> Weir Measurement Technique                             |
|  | Paper-IV  | Bioactive materials   |

**Physical Chemistry**

|   |                          |           |   |
|---|--------------------------|-----------|---|
| 1 | Mishra Ekta Jaiprakash   | Paper-I   | <b>Introduction:</b> Polymer Science, fundamental terms   |
|   |                          | Paper-II  | Variation of optical and magnetic properties of non material lwith size.  |
|   |                          | Paper-III | The Born–Oppenheimer approximation.   |
|   |                          | Paper-IV  | <b>Thermogravimetry (TG):</b> Principle and Instrumentation   |
| 2 | Khan Shehnaz Amirullah   | Paper-I   | Polymer Science historical outline, classification based on: the origin (natural, semi-synthetic, synthetic etc.) |
|   |                          | Paper-II  | Variation of optical and magnetic properties of non material lwith shape  |
|   |                          | Paper-III | The Born–Oppenheimer approximation  |
|   |                          | Paper-IV  | <b>Thermogravimetry (TG):</b> factors affecting thermo gravimetric curves   |
| 3 | Malik Sana A.            | Paper-I   | Polymer Science: the structure (linear, branched, network, hyper branched, dendrimer, ladder, cross linked, IPN), |
|   |                          | Paper-II  | Variation of optical and magnetic properties of non material with surface characteristics                         |
|   |                          | Paper-III | LCAO method-molecular orbital formation   |
|   |                          | Paper-IV  | <b>Thermogravimetry (TG):</b> Interpretation of thermo gravimetric curves   |
| 4 | Shukla Nikhil Pawankumar | Paper-I   | the type of atom in the main chain(homochain, heterochain)  |
|   |                          | Paper-II  | Variation of optical and magnetic properties of non material with impurities                                      |
|   |                          | Paper-III | LCAO method-molecular orbital formation   |
|   |                          | Paper-IV  | <b>Thermogravimetry (TG):</b> applications of thermo gravimetry   |

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M.Sc. Part II- Internal Mark List Sem-III (October-2021)

**Organic Chemistry**

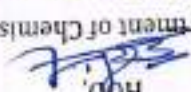
| Sr. No. | Name of Students               | Paper-I | Paper-II | Paper-III | Paper-IV |
|---------|--------------------------------|---------|----------|-----------|----------|
| 1.      | Harwan Saloni Sunil            | 34      | 30       | 32        | 31       |
| 2.      | Khan Nikhat Parveen Ahsanullah | 32      | 31       | 30        | 31       |
| 3.      | Khan Umar Farooq Khan          | 35      | 32       | 34        | 32       |
| 4.      | Sidique Mohammad Ahmad Abdul   | 33      | 33       | 34        | 33       |
| 5.      | Tiwari Sarvagya Kumar Arvind   | 32      | 30       | 29        | 30       |
| 6.      | Yadav Anish Anil               | 33      | 34       | 34        | 31       |
| 7.      | Yadav Dinesh Suresh            | 35      | 36       | 35        | 33       |

**Inorganic Chemistry**

| Sr. No. | Name of Students              | Paper-I | Paper-II | Paper-III | Paper-IV |
|---------|-------------------------------|---------|----------|-----------|----------|
| 1       | Beig Parvej Khwaja            | 27      | 33       | 28        | 25       |
| 2       | Kadam Pratik Prakash          | 28      | 29       | 29        | 22       |
| 3       | Khan Mohd Shaban Abdul Subhan | 32      | 25       | 32        | 23       |
| 4       | Mane Swapnali Yuvraj          | 26      | 28       | 30        | 22       |
| 5       | Manihar Rafiullah Shamsullah  | 16      | 21       | 16        | 21       |
| 6       | Qureshi Heena Jafar           | 27      | 32       | 34        | 21       |
| 7       | Singh Ashwini Anil            | 32      | 30       | 32        | 24       |

**Physical Chemistry**

| Sr. No. | Name of Students         | Paper-I | Paper-II | Paper-III | Paper-IV |
|---------|--------------------------|---------|----------|-----------|----------|
| 1       | Malik Sana Azimur Rehman | 34      | 31       | 29        | 34       |
| 2       | Shukla Nikhil Pawankumar | 27      | 28       | 27        | 31       |
| 3       | Khan Shehbaz Amirullah   | 25      | 30       | 27        | 33       |
| 4       | Mishra Ekta Jaiprakash   | 35      | 36       | 33        | 34       |

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Department of Chemistry

M.Sc. Part I

Semester II April 2022

| Sr. No | Name of Subject                             | Date and Time                  | Name of Examiner                           |
|--------|---|--------------------------------|--|
| 1      | Physical Chemistry: Course Code: PSCH 201   | 07/04/2022<br>9:00am to 2:00pm | Shri. Yatendra Yadav                       |
| 2      | Inorganic Chemistry: Course Code: PSCH 202  | 08/04/2022<br>9:00am to 2:00pm | Miss. Saira Siddique and<br>Ruqaiya Ansari |
| 3      | Organic Chemistry: Course Code: PSCH 203    | 11/04/2022<br>9:00am to 2:00pm | Dr. Dattatraya Bhangare                    |
| 4      | Analytical Chemistry: Course Code: PSCH 204 | 12/04/2022<br>9:00am to 2:00pm | Shri Pramod Vishwakarma                    |

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Department of Chemistry  
M. Sc. I Sem-II (Internal Presentation Topics) June-2022

| Sr. No. | Name of Students             | Paper No. | Topic For presentation   |
|---------|------------------------------|-----------|--|
| 1.      | Khan Iqra Azizullah          | Paper-I   | Maxwell equations, Maxwell thermodynamic Relations; its significance and applications to ideal gases.                    |
|         |                              | Paper-II  | Preparative methods: Chemical methods, Solvothermal,   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: rate and equilibrium constants                                     |
|         |                              | Paper-IV  | Laser as a source of radiation, Fibre optics   |
| 2.      | Khan Khushnuma Abdul Moid    | Paper-I   | Third law of Thermodynamics, Entropy change for a phase transition, absolute entropies                                   |
|         |                              | Paper-II  | Preparative methods: Solvothermal, Combustion synthesis,   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: reaction coordinate diagram, transition state (activated complex), |
|         |                              | Paper-IV  | Introduction of Fourier Transform  |
| 3.      | Khan Mehjabeen Mohammad      | Paper-I   | Joule Thomson experiment, Joule Thomson coefficient  |
|         |                              | Paper-II  | Preparative methods: Microwave, Co-precipitation,  |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: nature of activated complex, Hammond postulate,                    |
|         |                              | Paper-IV  | Derivation of Beer- Lambert's Law and its limitations  |
| 4.      | Khan Tabassum Ejaz           | Paper-I   | Joule Thomson coefficient in terms of van der Waals constants  |
|         |                              | Paper-II  | Preparative methods: Langmuir Blodgett (L-B) method,   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: Reactivity vs selectivity,   |
|         |                              | Paper-IV  | Factors affecting molecular absorption, types of transitions   |
| 5.      | Khan Zainab Parvin Noor      | Paper-I   | Standard molar entropies and their dependence on molecular mass and molecular structure                                  |
|         |                              | Paper-II  | Biological methods: Synthesis using microorganisms   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: Curtin-Hammett Principle,  |
|         |                              | Paper-IV  | factors affecting molecular absorption: pH, temperature,   |
| 6.      | Kumbhar Vaishnavi Shashikant | Paper-I   | Classical Mechanics, failure of classical mechanics  |
|         |                              | Paper-II  | Applications in the field of semiconductors,   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: Microscopic reversibility,   |
|         |                              | Paper-IV  | Factors affecting molecular absorption: solvent and effect of substituents.  |
| 7.      | Mahadeshwar Diksha Dilip     | Paper-I   | Need for Quantum Mechanics.  |
|         |                              | Paper-II  | Applications in the field of solar cells   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: Kinetic vs thermodynamic control of organic reactions.             |
|         |                              | Paper-IV  | Dual spectrometry – Introduction, Principle, Instrumentation and Applications  |
| 8.      | Mansoori Safa                | Paper-I   | Particle waves and Schrödinger wave equation   |



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|-----|--------------------------------|-----------|--|
|     | Banu Chand Farooq              | Paper-II  | Characterisation of Coordination compounds: Formation, thermal studies, Conductivity measurement   |
|     |                                | Paper-III | Ester hydrolysis: Classification   |
|     |                                | Paper-IV  | Applications of Ultraviolet and Visible spectroscopy: 1) On charge transfer absorption   |
| 9.  | Mansuri Md Jahangir Nizamuddin | Paper-I   | wave functions, properties of wave functions   |
|     |                                | Paper-II  | Characterisation of Coordination compounds: electronic spectral and magnetic measurements  |
|     |                                | Paper-III | Determining mechanism of a reaction: Product analysis,   |
|     |                                | Paper-IV  | Applications of Ultraviolet and Visible spectroscopy: Simultaneous spectroscopy  |
| 10. | Mishra Nishant Deopraakash     | Paper-I   | Normalization of wave functions,   |
|     |                                | Paper-II  | Characterisation of Coordination compounds: IR   |
|     |                                | Paper-III | Determining mechanism of a reaction: kinetic studies, use of isotopes (Kinetic isotope effect – primary and secondary kinetic isotope effect), |
|     |                                | Paper-IV  | Applications of Ultraviolet and Visible spectroscopy: Derivative Spectroscopy  |
| 11. | Narkar Ankita Sadashiv         | Paper-I   | Orthogonality of wave functions.   |
|     |                                | Paper-II  | Characterisation of Coordination compounds: NMR  |
|     |                                | Paper-III | Determining mechanism of a reaction: Detection and trapping of intermediates,  |
|     |                                | Paper-IV  | Oxidation number, rules for assigning oxidation number,  |
| 12. | Neman Kunal Eknath             | Paper-I   | Operators and their algebra  |
|     |                                | Paper-II  | Preparative methods: Chemical methods  |
|     |                                | Paper-III | Characterisation of Coordination compounds: ESR  |
|     |                                | Paper-IV  | Redox reaction in term of oxidation number, oxidizing and reducing agents,   |
| 13. | Pandhare Sudhir Bhanudas       | Paper-I   | Linear and Hermitian operators   |
|     |                                | Paper-II  | Microwave synthesis (discussion on principles, examples, merits and demerits are expected)   |
|     |                                | Paper-III | Determining mechanism of a reaction: crossover experiments and stereochemical evidence.  |
|     |                                | Paper-IV  | Equivalent weight of oxidizing and reducing agents, stoichiometry of redox titration   |
| 14. | Shaikh Abu Bakar Bakar Husain  | Paper-I   | Operators for the dynamic variables of a system such as, position, linear momentum, angular momentum   |
|     |                                | Paper-II  | Methods of preparation for inorganic solids: Ceramic method, precursor method, sol-gel method (applications in Biosensors),                    |
|     |                                | Paper-III | Determining mechanism of a reaction: stereochemical evidence.  |
|     |                                | Paper-IV  | Normality of a solution of a oxidizing / reducing agent and its relationship with molarity.  |
| 15. | Shaikh Arbaaz Siraj            | Paper-I   | Operators for the dynamic variables of a system such as, total energy, eigen functions, eigen values and eigen value equation                  |
|     |                                | Paper-II  | Layer structure [cadmium chloride and iodide ( $\text{CdCl}_2$ , $\text{CdI}_2$ )].  |
|     |                                | Paper-III | Acids and Bases: Factors affecting acidity and basicity:   |



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|     |                             |           |   |
|-----|-----------------------------|-----------|---|
|     |                             |           | Electronegativity   |
|     |                             | Paper-IV  | Calculations of pH of acids, bases, acidic and basic buffers.   |
| 16. | Shaikh Misba Sakhavat       | Paper-I   | Schrödinger wave equation as the eigen value equation of the Hamiltonian operator   |
|     |                             | Paper-II  | Electronic structure of solids and band theory  |
|     |                             | Paper-III | Acids and Bases: Factors affecting acidity and basicity: inductive effect   |
|     |                             | Paper-IV  | Calculations of pH of acids, bases, acidic and basic buffers.   |
| 17. | Shaikh Mohammed Ammar Ilyas | Paper-I   | Average value and the expectation value of a dynamic variable of the system   |
|     |                             | Paper-II  | Fermi level, K Space and Brillouin Zones.   |
|     |                             | Paper-III | Acids and Bases: Factors affecting acidity and basicity: resonance, bond strength, electrostatic effects,                 |
|     |                             | Paper-IV  | Solubility and solubility equilibria, effect of presence of common ion.   |
| 18. | Shaikh Tubba Ansar          | Paper-I   | Postulates of Quantum Mechanics   |
|     |                             | Paper-II  | Structures of Compounds of the type: AB [nickel arsenide (NiAs)], AB <sub>2</sub> [fluorite (CaF <sub>2</sub> )]          |
|     |                             | Paper-III | Acids and Bases: Factors affecting acidity and basicity: hybridization, aromaticity and solvation.                        |
|     |                             | Paper-IV  | Stoichiometry of chemical reactions, concept of kg mol, limiting reactant, theoretical and practical yield.               |
| 19. | Haikh Zainabalfiya Imtiaz   | Paper-I   | Schrodinger's Time independent wave equation  |
|     |                             | Paper-II  | Structures of Compounds of the type: AB [nickel arsenide (NiAs)].   |
|     |                             | Paper-III | Comparative study of acidity and basicity of organic compounds on the basis of pKa values.                                |
|     |                             | Paper-IV  | Calculations of ppm, ppb and dilution of the solutions, concept of mmol   |
| 20. | Singh Shrasti Manoj Kumar   | Paper-I   | Schrodinger's time dependent wave equation.   |
|     |                             | Paper-II  | Structures of Compounds of the type: AB <sub>2</sub> [fluorite (CaF <sub>2</sub> )]                                       |
|     |                             | Paper-III | Leveling effect and non-aqueous solvents.   |
|     |                             | Paper-IV  | Nomenclature of all eight mechanisms of acid and base catalyzed hydrolysis.   |
| 21. | Vhadade Amar Shripati       | Paper-I   | Application of quantum mechanics to the following systems: a) Free particle, wave function and energy of a free particle. |
|     |                             | Paper-II  | Structures of Compounds of the type: AB <sub>2</sub> rutile (TiO <sub>2</sub> ) structure                                 |
|     |                             | Paper-III | Acid and base catalysis – general and specific catalysis with examples.   |
|     |                             | Paper-IV  | Concentration of a solution based on volume and mass units.   |

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Department of Chemistry  
M. Sc. I Sem-II (Internal Marks) June-2022

| Sr. No. | Name of Students              | Paper-I | Paper-II | Paper-III | Paper-IV |
|---------|-------------------------------|---------|----------|-----------|----------|
| 1.      | Khan Iqra Azizullah           | 29      | 26       | 28        | 30       |
| 2.      | Khan Khushnara Abdul          | 32      | 26       | 31        | 30       |
| 3.      | Khan Mehjabeen Mohammad Shafi | 30      | 24       | 22        | 31       |
| 4.      | Khan Tabassum Ejaz            | 31      | 25       | 23        | 29       |
| 5.      | Khan Zainab Parvin Noor       | 33      | 26       | 26        | 30       |
| 6.      | Kumbhar Vaishnavi Shashikant  | 32      | 27       | 27        | 31       |
| 7.      | Mahadeshwar Diksha Dilip      | 27      | 26       | 24        | 32       |
| 8.      | Mansoori Safa Banu Chand      | 32      | 26       | 26        | 30       |
| 9.      | Ansuri Md Jahangir Nizamuddin | 30      | 26       | 25        | 29       |
| 10.     | Mishra Nishant Deoprasad      | 35      | 30       | 29        | 32       |
| 11.     | Narkar Ankita Sadashiv        | 27      | 24       | 27        | 28       |
| 12.     | Neman Kunal Eknath            | 32      | 23       | 25        | 30       |
| 13.     | Pandhare Sudhir Bhanudas      | 30      | 24       | 23        | 29       |
| 14.     | Shaikh Abu Bakar Bakar Husain | 31      | 25       | 22        | 30       |
| 15.     | Shaikh Arbaaz Siraj           | 27      | 22       | 27        | 30       |
| 16.     | Shaikh Misba Sakhat           | 30      | 26       | 28        | 31       |
| 17.     | Shaikh Mohammed Anwar Ilyas   | 31      | 26       | 25        | 31       |
| 18.     | Shaikh Tubha Ansar            | 30      | 25       | 24        | 30       |
| 19.     | Shaikh Zainabulfiya Imtiaz    | 35      | 32       | 27        | 31       |
| 20.     | Singh Shresthi Manoj Kumar    | 30      | 30       | 30        | 32       |
| 21.     | Vhadode Amar Shripati         | 26      | 26       | 23        | 30       |

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Department of Chemistry

M.Sc. Part II

Semester IV April 2022

Physical Chemistry

| Sr. No | Subject Code | Name of Subject   | Date and Time                   | Name of Examiner         |
|--------|--------------|---|---------------------------------|--------------------------|
| 1      | PSCHP401     | Chemistry: Polymer, Green, Biophysical and Applied.       | 18/04/2022<br>9:00am to 11:00am | Shri. Yatendra Yadav     |
| 2      | PSCHP402     | Material Science, Network And Irreversible Thermodynamics | 18/04/2022<br>12:00pm to 2:00pm | Shri. Rajeev Verma       |
| 3      | PSCHP403     | Symmetry & Spectroscopy                                   | 19/04/2022<br>9:00am to 11:00am | Miss. Azra Ansari        |
| 4      | PSCHP404     | Research Methodology                                      | 19/04/2022<br>12:00pm to 2:00pm | Shri. Pramod Vishwakarma |

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Department of Chemistry

M.Sc. Part II

Semester IV April 2022

Inorganic Chemistry

| Sr. No | Subject Code | Name of Subject                                  | Date and Time                   | Name of Examiner            |
|--------|--------------|--|---------------------------------|-----------------------------|
| 1      | PSCHP401     | Properties of Inorganic Solids and Group Theory. | 20/04/2022<br>9:00am to 11:00am | Miss. Saira Siddique        |
| 2      | PSCHP402     | Organometallic and main group Chemistry          | 20/04/2022<br>12:00pm to 2:00pm | Miss. Ruqaiya Ansari        |
| 3      | PSCHP403     | Instrumental methods in Inorganic Chemistry      | 21/04/2022<br>9:00am to 11:00am | Miss. Priyanka Sonar        |
| 4      | PSCHP404     | Research Methodology                             | 21/04/2022<br>12:00pm to 2:00pm | Shri. Pramod<br>Vishwakarma |

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Department of Chemistry

M.Sc. Part II

Semester IV April 2022

Organic Chemistry

| Sr. No | Subject Code | Name of Subject                             | Date and Time                   | Name of Examiner         |
|--------|--------------|---|---------------------------------|--------------------------|
| 1      | PSCHO401     | Theoretical organic chemistry-II            | 22/04/2022<br>9:00am to 11:00am | Miss. Pratiksha Borase   |
| 2      | PSCHO402     | Synthetic organic chemistry-II              | 22/04/2022<br>12:00pm to 2:00pm | Dr. Kiran Taksande       |
| 3      | PSCHO403     | Natural products and heterocyclic chemistry | 23/04/2022<br>9:00am to 11:00am | Dr. Dattatraya Bhangare  |
| 4      | PSCHO404     | Research Methodology                        | 23/04/2022<br>12:00pm to 2:00pm | Shri. Pramod Vishwakarma |

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**M.Sc. II Sem-IV, Topic For Presentation (Internal Exam)- June-2022**

Organic Chemistry

| Sr. No. | Name of Students                     | Paper No. | Topic For presentation   |
|---------|--------------------------------------|-----------|--|
| 1.      | Harwani Saloni<br>Sumil              | Paper-I   | Structural effects and reactivity: Linear free energy relationship (LFER) in determination of organic reaction mechanism.                |
|         |                                      | Paper-II  | <b>Protecting groups in Organic Synthesis:</b> Protection and deprotection of the hydroxyl.  |
|         |                                      | Paper-III | <b>Steroids:</b> General structure, classification   |
|         |                                      | Paper-IV  | History and evolution of cheminformatics   |
| 2.      | Khan Nikhat<br>Parveen<br>Ahsanullah | Paper-I   | The Hammett equation, substituent constants  |
|         |                                      | Paper-II  | <b>Protecting groups in Organic Synthesis:</b> Protection and deprotection of the carbonyl   |
|         |                                      | Paper-III | <b>Steroids:</b> Occurrence, biological role, important structural and stereochemical features of: corticosteroids.                      |
|         |                                      | Paper-IV  | Use of Cheminformatics.  |
| 3.      | Khan Umar<br>Farooq Khan             | Paper-I   | Theories of substituent effects, interpretation of $\sigma$ -values  |
|         |                                      | Paper-II  | <b>Protecting groups in Organic Synthesis:</b> Protection and deprotection of the amino.   |
|         |                                      | Paper-III | <b>Steroids:</b> Occurrence, biological role, important structural and stereochemical features of: steroidal hormones                    |
|         |                                      | Paper-IV  | Prospects of cheminformatics   |
| 4.      | Sidique Mohamad<br>Ahmad Abdul       | Paper-I   | Theories of substituent effects, interpretation of $\sigma$ -values  |
|         |                                      | Paper-II  | <b>Protecting groups in Organic Synthesis:</b> Protection and deprotection of the carboxyl functional groups and its applications.       |
|         |                                      | Paper-III | <b>Steroids:</b> Occurrence, biological role, important structural and stereochemical features of: steroidal alkaloids.                  |
|         |                                      | Paper-IV  | Molecular modeling and structure elucidation.  |
| 5.      | Tiwari Sarvagya<br>Kumar Arvind      | Paper-I   | Uses of Hammett equation, deviations from Hammett equation.  |
|         |                                      | Paper-II  | <b>Concept of umpolung (Reversal of polarity):</b> Generation of acyl anion equivalent using 1,3-dithianes, methyl thiomethyl sulfoxides |
|         |                                      | Paper-III | <b>Steroids:</b> Occurrence, biological role, important structural and stereochemical features of: sterols and bile acids.               |
|         |                                      | Paper-IV  | <b>Representation of molecules and chemical reactions:</b> Nomenclature, Different types of notations                                    |
| 6.      | Yadav Anish Anil                     | Paper-I   | Neighbouring group participation: Mechanism and effects of anchimeric assistance   |
|         |                                      | Paper-II  | Multicomponent reactions: Strecker Synthesis, Ugi 4CC,   |
|         |                                      | Paper-III | General structural features, occurrence, biological importance and applications of: carotenoids  |



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|----|---------------------|-----------|--|
| 7. | Yadav Dinesh Suresh | Paper-IV  | Shikimic Acid pathway: Biosynthesis of shikimic acid   |
|    |                     | Paper-I   | NGP by unshared/ lone pair electrons, $\pi$ -electrons, aromatic rings                           |
|    |                     | Paper-II  | Multicomponent reactions: Biginelli synthesis, Hantzsch synthesis                                |
|    |                     | Paper-III | General structural features, occurrence, biological importance and applications of: anthocyanins |
|    |                     | Paper-IV  | Shikimic Acid pathway: Biosynthesis of aromatic amino acids, cinnamic acid and its derivatives   |

**Inorganic Chemistry**

|   |                      |           |  |
|---|----------------------|-----------|--|
| 1 | Beig Parvej Khwaja   | Paper-I   | <b>Electrical properties of solids:</b><br>(i) Conductivity: Solid Electrolytes; Fast Ion                                      |
|   |                      | Paper-II  | <b>Organometallic Chemistry:</b><br>(a) Metal-Metal Bonding and Metal Clusters   |
|   |                      | Paper-III | <b>Infrared spectroscopy:</b> Fundamental modes of vibrations, selection rules   |
|   |                      | Paper-IV  | Primary, Secondary and Tertiary sources  |
| 2 | Kadam Pratik Prakash | Paper-I   | Conductors; Mechanism of Conductivity; Hopping Conduction.   |
|   |                      | Paper-II  | <b>Organometallic Chemistry:</b> Isolobal Analogy  |
|   |                      | Paper-III | <b>Infrared spectroscopy:</b> IR absorption bands of metal - donor atom  |
|   |                      | Paper-IV  | Journal abbreviations, abstracts, current titles, reviews  |
| 3 | Mane Swapnali Yuvraj | Paper-I   | <b>Other Electrical Properties:</b> Thomson and Seebeck Effects;   |
|   |                      | Paper-II  | <b>Organometallic Chemistry:</b> Organo Palladium Complexes (preparations, properties and applications.)                       |
|   |                      | Paper-III | <b>Infrared spectroscopy:</b> effect of complexation on the IR spectrum of ligands formations on the IR of ligands like $NH_3$ |
|   |                      | Paper-IV  | monographs, dictionaries, textbooks, current contents  |
| 4 | Qureshi Heena Jafar  | Paper-I   | Thermocouples and their Applications;  |
|   |                      | Paper-II  | <b>Organometallic Chemistry:</b> Organo Platinum Complexes (preparations, properties and applications)                         |
|   |                      | Paper-III | <b>Infrared spectroscopy:</b> effect of complexation on the IR spectrum of ligands formations on the IR of ligands like $CN^-$ |
|   |                      | Paper-IV  | Introduction to Chemical Abstracts and Beilstein, Subject Index other Indices with examples.                                   |
| 5 | Singh Ashwini Anil   | Paper-I   | Hall Effect; Dielectric, Ferroelectric,  |
|   |                      | Paper-II  | Substance Index, Author Index, Formula Index   |
|   |                      | Paper-III | <b>Infrared spectroscopy:</b> effect of complexation on the IR spectrum of ligands formations on the IR of ligands like CO     |

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|---|------------------------|------------------|--|
|   |                        | <b>Paper-IV</b>  | Web sources, E-journals, Journal access, TOC alerts                          |
| 6 | Khan Mohd Shaban Abdul | <b>Paper-I</b>   | Hall Effect; Piezoelectric and Pyroelectric                                  |
|   |                        | <b>Paper-II</b>  | <b>Organometallic Chemistry:</b> (a) Metal-Metal Bonding and Metal Clusters  |
|   |                        | <b>Paper-III</b> | <b>Raman spectroscopy:</b> Raman spectroscopy for diatomic molecules         |
|   |                        | <b>Paper-IV</b>  | Hot articles, Citation Index, Impact factor, H-index, E-consortium           |
| 7 | Manihar Rafiullah S.   | <b>Paper-I</b>   | Materials and their Inter-relationships                                      |
|   |                        | <b>Paper-II</b>  | Catalysis-Homogenous and Heterogenous Catalysis: Comparison.                 |
|   |                        | <b>Paper-III</b> | Determination of molecular structures like diatomic and triatomic molecules. |
|   |                        | <b>Paper-IV</b>  | Bioactive materials  |

**Physical Chemistry**

|   |                          |                  |  |
|---|--------------------------|------------------|--|
| 1 | Mishra Ekta Jaiprakash   | <b>Paper-I</b>   | <b>Polymers in solid state</b> – Transitions (glass transition and crystalline melting temperature)                          |
|   |                          | <b>Paper-II</b>  | <b>Metals and alloys:</b> Solidification of metals and alloys-homogeneous and heterogeneous nucleation<br>Growth of crystals |
|   |                          | <b>Paper-III</b> | Recapitulation: point groups, character tables.  |
|   |                          | <b>Paper-IV</b>  | weighted linear case, analysis of residuals  |
| 2 | Khan Shehnaz Amirullah   | <b>Paper-I</b>   | <b>Polymers in solid state</b> crystalline behaviour, factors affecting crystallinity  |
|   |                          | <b>Paper-II</b>  | <b>Metals and alloys:</b> Variation of optical and magnetic properties of non material with shape.                           |
|   |                          | <b>Paper-III</b> | Recapitulation: character tables   |
|   |                          | <b>Paper-IV</b>  | fitting of linear equations, simple linear cases   |
| 3 | Malik Suna A.            | <b>Paper-I</b>   | <b>Polymers in solid state:</b> polymer blends and Alloys.   |
|   |                          | <b>Paper-II</b>  | <b>Metals and alloys:</b> growth of silicon single crystal   |
|   |                          | <b>Paper-III</b> | Reduction formula, application of reduction formula to vibrational modes of water molecule                                   |
|   |                          | <b>Paper-IV</b>  | Correlation and regression, curve fitting  |
| 4 | Shukla Nikhil Pawankumar | <b>Paper-I</b>   | <b>Identification and characterization of polymers:</b> Chemical analysis- End group analysis                                |
|   |                          | <b>Paper-II</b>  | <b>Metals and alloys:</b> growth of silicon single crystal   |
|   |                          | <b>Paper-III</b> | Reduction formula, application of reduction formula to vibrational modes of water molecule                                   |
|   |                          | <b>Paper-IV</b>  | Chemometrics, Analysis of Variance (ANOVA)   |

  
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**M.Sc. Part II- Internal Mark List Sem-IV June 2022(internal Marks)**

**Organic Chemistry**

| Sr. No. | Name of Students               | Paper-I | Paper-II | Paper-III | Paper-IV |
|---------|--------------------------------|---------|----------|-----------|----------|
| 1.      | Harwani Saloni Sunil           | 34      | 30       | 32        | 31       |
| 2.      | Khan Nikhat Parveen Ahsanullah | 32      | 31       | 30        | 31       |
| 3.      | Khan Umar Farooq Khan          | 35      | 32       | 34        | 32       |
| 4.      | Sidique Mohamad Ahmad Abdul    | 33      | 33       | 34        | 33       |
| 5.      | Tiwari Sarvagya Kumar Arvind   | 32      | 30       | 29        | 30       |
| 6.      | Yadav Anish Anil               | 33      | 34       | 34        | 31       |
| 7.      | Yadav Dinesh Suresh            | 35      | 36       | 35        | 33       |

**Inorganic Chemistry**

| Sr. No. | Name of Students              | Paper-I | Paper-II | Paper-III | Paper-IV |
|---------|-------------------------------|---------|----------|-----------|----------|
| 1       | Beig Parvej Khwaja            | 27      | 33       | 28        | 25       |
| 2       | Kadam Pratik Prakash          | 28      | 29       | 29        | 22       |
| 3       | Khan Mohd Shaban Abdul Subhan | 32      | 25       | 32        | 23       |
| 4       | Mane Swapnali Yuvraj          | 26      | 28       | 30        | 22       |
| 5       | Manihar Rafiullah Shamsullah  | 16      | 21       | 16        | 21       |
| 6       | Qureshi Heena Jafar           | 27      | 32       | 34        | 21       |
| 7       | Singh Ashwini Anil            | 32      | 30       | 32        | 24       |

**Physical Chemistry**

| Sr. No. | Name of Students         | Paper-I | Paper-II | Paper-III | Paper-IV |
|---------|--------------------------|---------|----------|-----------|----------|
| 1       | Malik Sana Azimurrehman  | 34      | 31       | 29        | 34       |
| 2       | Shukla Nikhil Pawankumar | 27      | 28       | 27        | 31       |
| 3       | Khan Shehnaz Amirullah   | 25      | 30       | 27        | 33       |
| 4       | Mishra Ekta Jaiprakash   | 35      | 36       | 33        | 34       |

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Department of Chemistry

M.Sc. Part I

Semester I December 2022

| Sr. No | Name of Subject                             | Date and Time                      | Name of Examiner                           |
|--------|---|------------------------------------|--|
| 1      | Physical Chemistry: Course Code: PSCH 101   | 19.12.2022<br>(09:00am to 05:00pm) | Shri. Yatendra Yadav                       |
| 2      | Inorganic Chemistry: Course Code: PSCH 102  | 20.12.2022<br>(09:00am to 05:00pm) | Miss. Saira Siddique and<br>Ruqaiya Ansari |
| 3      | Organic Chemistry: Course Code: PSCH 103    | 21.12.2022<br>(09:00am to 05:00pm) | Dr. Dattatraya Bhangare                    |
| 4      | Analytical Chemistry: Course Code: PSCH 104 | 22.12.2022<br>(09:00am to 05:00pm) | Shri Pramod Vishwakarma                    |

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Department of Chemistry  
M. Sc. I (Internal Presentation Topics) December-2022

| Sr. No. | Name of Students             | Paper No. | Topic For presentation   |
|---------|------------------------------|-----------|--|
| 1.      | Behera Purnima<br>Laxmidhar  | Paper-I   | Maxwell equations, Maxwell thermodynamic Relations; its significance and applications to ideal gases.                    |
|         |                              | Paper-II  | Preparative methods: Chemical methods, Solvothermal,   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: rate and equilibrium constants                                     |
|         |                              | Paper-IV  | Laser as a source of radiation, Fibre optics   |
| 2.      | Gupta Rohit<br>Kumar P.      | Paper-I   | Third law of Thermodynamics, Entropy change for a phase transition, absolute entropies                                   |
|         |                              | Paper-II  | Preparative methods: Solvothermal, Combustion synthesis,   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: reaction coordinate diagram, transition state (activated complex), |
|         |                              | Paper-IV  | Introduction of Fourier Transform.   |
| 3.      | Gupta Shivangi<br>Rajendra   | Paper-I   | Joule Thomson experiment, Joule Thomson coefficient  |
|         |                              | Paper-II  | Preparative methods: Microwave, Co-precipitation,  |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: nature of activated complex, Hammond postulate,                    |
|         |                              | Paper-IV  | Derivation of Beer- Lambert's Law and its limitations  |
| 4.      | Khan Hadiya<br>Khatun Abusad | Paper-I   | Joule Thomson coefficient in terms of van der Waals constants  |
|         |                              | Paper-II  | Preparative methods: Langmuir Blodgett (L-B) method,   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: Reactivity vs selectivity,   |
|         |                              | Paper-IV  | Factors affecting molecular absorption, types of transitions   |
| 5.      | Khan Parveen<br>Amjad Ali    | Paper-I   | Standard molar entropies and their dependence on molecular mass and molecular structure                                  |
|         |                              | Paper-II  | Biological methods: Synthesis using microorganisms   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: Curtin-Hammett Principle,  |
|         |                              | Paper-IV  | factors affecting molecular absorption: pH, temperature,   |
| 6.      | Khan Shamina<br>Bano Ansar   | Paper-I   | Classical Mechanics, failure of classical mechanics  |
|         |                              | Paper-II  | Applications in the field of semiconductors,   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: Microscopic reversibility,   |
|         |                              | Paper-IV  | Factors affecting molecular absorption: solvent and effect of substituents.  |
| 7.      | Memon Aisha<br>Salim         | Paper-I   | Need for Quantum Mechanics.  |
|         |                              | Paper-II  | Applications in the field of solar cells   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: Kinetic vs thermodynamic control of organic reactions,             |
|         |                              | Paper-IV  | Dual spectrometry – Introduction, Principle, Instrumentation and Applications  |
| 8.      | Mishra Ankit                 | Paper-I   | Particle waves and Schrödinger wave equation   |



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|     |                                   |           |  |
|-----|-----------------------------------|-----------|--|
|     | Kumar Anil                        | Paper-II  | Characterisation of Coordination compounds: Formation, thermal studies, Conductivity measurement   |
|     |                                   | Paper-III | Ester hydrolysis: Classification   |
|     |                                   | Paper-IV  | Applications of Ultraviolet and Visible spectroscopy: 1) On charge transfer absorption   |
| 9.  | Mishra Shishir<br>Santosh         | Paper-I   | wave functions, properties of wave functions   |
|     |                                   | Paper-II  | Characterisation of Coordination compounds: electronic spectral and magnetic measurements  |
|     |                                   | Paper-III | Determining mechanism of a reaction: Product analysis,   |
|     |                                   | Paper-IV  | Applications of Ultraviolet and Visible spectroscopy: Simultaneous spectroscopy  |
| 10. | Mishra Utkarsh<br>Narendra        | Paper-I   | Normalization of wave functions,   |
|     |                                   | Paper-II  | Characterisation of Coordination compounds: IR   |
|     |                                   | Paper-III | Determining mechanism of a reaction: kinetic studies, use of isotopes (Kinetic isotope effect – primary and secondary kinetic isotope effect). |
|     |                                   | Paper-IV  | Applications of Ultraviolet and Visible spectroscopy: Derivative Spectroscopy  |
| 11. | Khan Roshan<br>Jahan<br>Kyamuddin | Paper-I   | Orthogonality of wave functions.   |
|     |                                   | Paper-II  | Characterisation of Coordination compounds: NMR  |
|     |                                   | Paper-III | Determining mechanism of a reaction: Detection and trapping of intermediates,  |
|     |                                   | Paper-IV  | Oxidation number, rules for assigning oxidation number,  |
| 12. | Shaikh Afrin<br>Mushtaq           | Paper-I   | Operators and their algebra  |
|     |                                   | Paper-II  | Preparative methods: Chemical methods  |
|     |                                   | Paper-III | Characterisation of Coordination compounds: ESR  |
|     |                                   | Paper-IV  | Redox reaction in term of oxidation number, oxidizing and reducing agents,   |
| 13. | Shaikh Ayesha<br>Riyaz            | Paper-I   | Linear and Hermitian operators   |
|     |                                   | Paper-II  | Microwave synthesis (discussion on principles, examples, merits and demerits are expected)   |
|     |                                   | Paper-III | Determining mechanism of a reaction: crossover experiments and stereochemical evidence.  |
|     |                                   | Paper-IV  | Equivalent weight of oxidizing and reducing agents, stoichiometry of redox titration   |
| 14. | Shaikh Salim<br>Dilbahar          | Paper-I   | Operators for the dynamic variables of a system such as, position, linear momentum, angular momentum   |
|     |                                   | Paper-II  | Methods of preparation for inorganic solids: Ceramic method, precursor method, sol-gel method (applications in Biosensors).                    |
|     |                                   | Paper-III | Determining mechanism of a reaction: stereochemical evidence.  |
|     |                                   | Paper-IV  | Normality of a solution of a oxidizing / reducing agent and its relationship with molarity.  |
| 15. | Shaikh<br>Mohammad<br>Sufiyan M.  | Paper-I   | Operators for the dynamic variables of a system such as, total energy, eigen functions, eigen values and eigen value equation                  |
|     |                                   | Paper-II  | Layer structure [cadmium chloride and iodide ( $\text{CdCl}_2$ , $\text{CdI}_2$ )].  |



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|     |                           |           |   |
|-----|---------------------------|-----------|---|
|     |                           | Paper-III | Acids and Bases: Factors affecting acidity and basicity: Electronegativity  |
|     |                           | Paper-IV  | Calculations of pH of acids, bases, acidic and basic buffers.   |
| 16. | Shingada Chandresh Radaka | Paper-I   | Schrödinger wave equation as the eigen value equation of the Hamiltonian operator   |
|     |                           | Paper-II  | Electronic structure of solids and band theory  |
|     |                           | Paper-III | Acids and Bases: Factors affecting acidity and basicity: inductive effect   |
|     |                           | Paper-IV  | Calculations of pH of acids, bases, acidic and basic buffers.   |
|     |                           |           |   |
| 17. | Siddiqui Hajra Dilshad A. | Paper-I   | Average value and the expectation value of a dynamic variable of the system   |
|     |                           | Paper-II  | Fermi level, K Space and Brillouin Zones.   |
|     |                           | Paper-III | Acids and Bases: Factors affecting acidity and basicity: resonance, bond strength, electrostatic effects,                 |
|     |                           | Paper-IV  | Solubility and solubility equilibria, effect of presence of common ion.   |
| 18. | Singh Vinay Ramesh        | Paper-I   | Postulates of Quantum Mechanics   |
|     |                           | Paper-II  | Structures of Compounds of the type: AB [nickel arsenide (NiAs)], AB <sub>2</sub> [fluorite (CaF <sub>2</sub> )]          |
|     |                           | Paper-III | Acids and Bases: Factors affecting acidity and basicity: hybridization, aromaticity and solvation.                        |
|     |                           | Paper-IV  | Stoichiometry of chemical reactions, concept of kg mol, limiting reactant, theoretical and practical yield.               |
| 19. | Tiwari Anjali Rajesh      | Paper-I   | Schrodinger's Time independent wave equation  |
|     |                           | Paper-II  | Structures of Compounds of the type: AB [nickel arsenide (NiAs)],   |
|     |                           | Paper-III | Comparative study of acidity and basicity of organic compounds on the basis of pK <sub>a</sub> values.                    |
|     |                           | Paper-IV  | Calculations of ppm, ppb and dilution of the solutions, concept of mmol   |
| 20. | Yadav Mansi Keshav        | Paper-I   | Schrodinger's time dependent wave equation.   |
|     |                           | Paper-II  | Structures of Compounds of the type: AB <sub>2</sub> [fluorite (CaF <sub>2</sub> )]                                       |
|     |                           | Paper-III | Leveling effect and non-aqueous solvents.   |
|     |                           | Paper-IV  | Nomenclature of all eight mechanisms of acid and base catalyzed hydrolysis.   |
| 21. | Yadav Neha Pramod         | Paper-I   | Application of quantum mechanics to the following systems: a) Free particle, wave function and energy of a free particle. |
|     |                           | Paper-II  | Structures of Compounds of the type: AB <sub>2</sub> rutile (TiO <sub>2</sub> ) structure                                 |
|     |                           | Paper-III | Acid and base catalysis – general and specific catalysis with examples.   |
|     |                           | Paper-IV  | Concentration of a solution based on volume and mass units.   |

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Department of Chemistry  
M. Sc. I, Sem-I (Internal Marks) December-2022

| Sr. No. | Name Of Students           | Paper-I | Paper-II | Paper-III | Paper-IV |
|---------|----------------------------|---------|----------|-----------|----------|
| 1.      | Behera Purnima Laxmidhar   | 29      | 24       | 25        | 24       |
| 2.      | Gupta Rohit Kumar P.       | 26      | 22       | 20        | 24       |
| 3.      | Gupta Shivanji Rajendra    | 29      | 24       | 30        | 24       |
| 4.      | Khan Hadiya Khatoon Abusad | 31      | 33       | 32        | 25       |
| 5.      | Khan Parveen Amjad Ali     | 28      | 30       | 28        | 26       |
| 6.      | Khan Shamina Bano Ansar    | 28      | 26       | 23        | 25       |
| 7.      | Memon Aisha Salim          | 34      | 35       | 36        | 33       |
| 8.      | Mishra Ankit Kumar Anil    | 24      | 28       | 26        | 29       |
| 9.      | Mishra Shishir Santosh     | 24      | 25       | 25        | 30       |
| 10.     | Mishra Vikash Narendra     | 28      | 24       | 22        | 30       |
| 11.     | Roshan Jahan Kyanuddin     | 30      | 30       | 31        | 24       |
| 12.     | Shaikh Afrin Mushtaq       | 25      | 29       | 28        | 26       |
| 13.     | Shaikh Ayesha Riyaz        | 28      | 30       | 25        | 24       |
| 14.     | Shaikh Salim Dilbahar      | 27      | 21       | 20        | 30       |
| 15.     | Shaikh Mohammad Suifan M.  | 29      | 21       | 22        | 28       |
| 16.     | Shingada Chandresh Radaka  | 28      | 22       | 21        | 29       |
| 17.     | Siddiqui Hajra Dilshad A.  | 24      | 25       | 24        | 24       |
| 18.     | Singh Vinay Ramesh         | 30      | 25       | 26        | 27       |
| 19.     | Tiwari Anjali Rajesh       | 24      | 28       | 26        | 28       |
| 20.     | Yadav Mansi Keshav         | 24      | 28       | 26        | 28       |
| 21.     | Yadav Neha Pramod          | 27      | 25       | 23        | 24       |

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Department of Chemistry

M.Sc. Part II

Semester IV April 2022

Organic Chemistry

| Sr. No | Subject Code | Name of Subject                             | Date and Time                   | Name of Examiner         |
|--------|--------------|---|---------------------------------|--------------------------|
| 1      | PSCHO401     | Theoretical organic chemistry-II            | 22/04/2022<br>9:00am to 11:00am | Miss. Pratiksha Borase   |
| 2      | PSCHO402     | Synthetic organic chemistry-II              | 22/04/2022<br>12:00pm to 2:00pm | Dr. Kiran Taksande       |
| 3      | PSCHO403     | Natural products and heterocyclic chemistry | 23/04/2022<br>9:00am to 11:00am | Dr. Dattatraya Bhangare  |
| 4      | PSCHO404     | Research Methodology                        | 23/04/2022<br>12:00pm to 2:00pm | Shri. Pramod Vishwakarma |

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Department of Chemistry

M.Sc. Part II

Semester IV April 2022

Inorganic Chemistry

| Sr. No | Subject Code | Name of Subject                                  | Date and Time                   | Name of Examiner            |
|--------|--------------|--|---------------------------------|-----------------------------|
| 1      | PSCHP401     | Properties of Inorganic Solids and Group Theory. | 20/04/2022<br>9:00am to 11:00am | Miss. Saira Siddique        |
| 2      | PSCHP402     | Organometallic and main group Chemistry          | 20/04/2022<br>12:00pm to 2:00pm | Miss. Ruqaiya Ansari        |
| 3      | PSCHP403     | Instrumental methods in Inorganic Chemistry      | 21/04/2022<br>9:00am to 11:00am | Miss. Priyanka Sonar        |
| 4      | PSCHP404     | Research Methodology                             | 21/04/2022<br>12:00pm to 2:00pm | Shri. Pramod<br>Vishwakarma |

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Department of Chemistry

M.Sc. Part II

Semester IV April 2022

Physical Chemistry

| Sr. No | Subject Code | Name of Subject   | Date and Time                   | Name of Examiner         |
|--------|--------------|---|---------------------------------|--------------------------|
| 1      | PSCHP401     | Chemistry: Polymer, Green, Biophysical and Applied.       | 18/04/2022<br>9:00am to 11:00am | Shri. Yatendra Yadav     |
| 2      | PSCHP402     | Material Science, Network And Irreversible Thermodynamics | 18/04/2022<br>12:00pm to 2:00pm | Shri. Rajeev Verma       |
| 3      | PSCHP403     | Symmetry & Spectroscopy                                   | 19/04/2022<br>9:00am to 11:00am | Miss. Azra Ansari        |
| 4      | PSCHP404     | Research Methodology                                      | 19/04/2022<br>12:00pm to 2:00pm | Shri. Pramod Vishwakarma |

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Department of Chemistry

M.Sc. Part I

Semester II April 2022

| Sr. No | Name of Subject                             | Date and Time                  | Name of Examiner                           |
|--------|---|--------------------------------|--|
| 1      | Physical Chemistry: Course Code: PSCH 201   | 07/04/2022<br>9:00am to 2:00pm | Shri. Yatendra Yadav                       |
| 2      | Inorganic Chemistry: Course Code: PSCH 202  | 08/04/2022<br>9:00am to 2:00pm | Miss. Saira Siddique and<br>Ruqaiya Ansari |
| 3      | Organic Chemistry: Course Code: PSCH 203    | 11/04/2022<br>9:00am to 2:00pm | Dr. Dattatraya Bhangare                    |
| 4      | Analytical Chemistry: Course Code: PSCH 204 | 12/04/2022<br>9:00am to 2:00pm | Shri Pramod Vishwakarma                    |

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**Topic for internal Presentation December-2022**

Organic Chemistry

| Sr. No. | Name of Students             | Paper No. | Topic For presentation  |
|---------|------------------------------|-----------|---|
| 1.      | Khan Iqra Azizullah          | Paper-I   | Organic reactive intermediates, methods of generation, structure, stability and important reactions involving carbocations. |
|         |                              | Paper-II  | Mukaiyama esterification, Mitsunobu reaction  |
|         |                              | Paper-III | Introduction to naturally occurring sugars: Deoxysugars.  |
|         |                              | Paper-IV  | Primary and secondary metabolites and the building blocks   |
| 2.      | Khan Khushnuma Abdul         | Paper-I   | Organic reactive intermediates, methods of generation, structure, stability and important reactions involving nitrenes.     |
|         |                              | Paper-II  | Darzen's Glycidic Ester synthesis, Ritter reaction  |
|         |                              | Paper-III | Introduction to naturally occurring sugars: Aminosugars.  |
|         |                              | Paper-IV  | General pathway of amino acid biosynthesis.   |
| 3.      | Mansoori Safa Bahu Chand     | Paper-I   | Organic reactive intermediates, methods of generation, structure, stability and important reactions involving carbenes.     |
|         |                              | Paper-II  | Yamaguchi esterification, Peterson olefination  |
|         |                              | Paper-III | Acetate pathway: Biosynthesis of malonylCoA   |
|         |                              | Paper-IV  | Derivation of Beer- Lambert's Law and its limitations   |
| 4.      | Khan Zainab Parvin Noor      | Paper-I   | Organic reactive intermediates, methods of generation, structure, stability and important reactions involving arynes.       |
|         |                              | Paper-II  | Domino reactions: Characteristics   |
|         |                              | Paper-III | Introduction to naturally occurring sugars: branched sugars.  |
|         |                              | Paper-IV  | Biosynthesis of saturated fatty acids.  |
| 5.      | Kumbhar Vaishnavi Shashikant | Paper-I   | Organic reactive intermediates, methods of generation, structure, stability and important reactions involving ketenes.      |
|         |                              | Paper-II  | Nazarov cyclization   |
|         |                              | Paper-III | Structure elucidation of lactose and Dglucosamine   |
|         |                              | Paper-IV  | Biosynthesis of prostaglandins from arachidonic acid.   |
| 6.      | Mishra Nishant Deoprasaksh   | Paper-I   | Neighbouring group participation: Mechanism and effects of anchimeric assistance  |
|         |                              | Paper-II  | Multicomponent reactions: Strecker Synthesis, Ugi 4CC,  |
|         |                              | Paper-III | General structural features, occurrence, biological importance and applications of: carotenoids                             |

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|    |                              |                  |  |
|----|------------------------------|------------------|--|
| 7. | Shaikh Zainabaliya<br>Imtiaz | <b>Paper-IV</b>  | Shikimic Acid pathway: Biosynthesis of shikimic acid   |
|    |                              | <b>Paper-I</b>   | NGP by unshared/ lone pair electrons, $\pi$ -electrons, aromatic rings                           |
|    |                              | <b>Paper-II</b>  | Multicomponent reactions: Biginelli synthesis, Hantzsch synthesis                                |
|    |                              | <b>Paper-III</b> | General structural features, occurrence, biological importance and applications of: anthocyanins |
|    |                              | <b>Paper-IV</b>  | Shikimic Acid pathway: Biosynthesis of aromatic amino acids, cinnamic acid and its derivatives   |

**Inorganic Chemistry**

|   |                                   |                  |   |
|---|-----------------------------------|------------------|---|
| 1 | Khan Mehjabeen<br>Mohammad Shafi  | <b>Paper-I</b>   | Structures of AB type compounds (PbO and CuO)   |
|   |                                   | <b>Paper-II</b>  | Coordination geometry of the metal ion and functions.   |
|   |                                   | <b>Paper-III</b> | X-Ray Diffraction: Bragg Condition; Miller Indices  |
|   |                                   | <b>Paper-IV</b>  | Handling of Hazardous Materials   |
| 2 | Mahadeshwar<br>Diksha Dilip       | <b>Paper-I</b>   | Structures of AB <sub>2</sub> type (β-cristobalite, CaC <sub>2</sub> and Cs <sub>2</sub> O)   |
|   |                                   | <b>Paper-II</b>  | Coordination geometry of the metal ion and functions.   |
|   |                                   | <b>Paper-III</b> | Miller Indices; Laue Method   |
|   |                                   | <b>Paper-IV</b>  | Toxic Materials Various types of toxins and their effects on humans   |
| 3 | Mansuri Md<br>Jahangir Nizamuddin | <b>Paper-I</b>   | Structures of A <sub>2</sub> B <sub>3</sub> type (Cr <sub>2</sub> O <sub>3</sub> and Bi <sub>2</sub> O <sub>3</sub> )               |
|   |                                   | <b>Paper-II</b>  | Coordination geometry of the metal ion and functions.   |
|   |                                   | <b>Paper-III</b> | Zn in biological systems: Carbonic anhydrase.   |
|   |                                   | <b>Paper-IV</b>  | Toxic Materials Various types of toxins and their effects on humans   |
| 4 | Neman Kunal<br>Eknath             | <b>Paper-I</b>   | Structures of B <sub>2</sub> (ReO <sub>3</sub> , Li <sub>3</sub> N)   |
|   |                                   | <b>Paper-II</b>  | Zn in biological systems: protolytic enzymes.   |
|   |                                   | <b>Paper-III</b> | Bragg Method; Debye Scherrer Method of X-Ray Structural Analysis of Crystals.   |
|   |                                   | <b>Paper-IV</b>  | Explosives and Inflammable Materials  |
| 5 | Pandhare Sudhir<br>Bhanudas       | <b>Paper-I</b>   | Structures of ABO <sub>3</sub> type, relation between ReO <sub>3</sub> and perovskite BaTiO <sub>3</sub> and its polymorphic forms. |
|   |                                   | <b>Paper-II</b>  | Zn in biological systems: carboxy peptidase, Zinc finger.   |
|   |                                   | <b>Paper-III</b> | <b>Electron Diffraction:</b> Scattering of electron   |
|   |                                   | <b>Paper-IV</b>  | Explosives and Inflammable Materials  |
| 6 | Singh Shrasti Manoj<br>Kumar      | <b>Paper-I</b>   | Structures of ABO <sub>3</sub> type, relation between ReO <sub>3</sub> and perovskite BaTiO <sub>3</sub> and its Oxide bronzes.     |
|   |                                   | <b>Paper-II</b>  | Role of metal ions in biological electron transfer processes: iron sulphur proteins   |
|   |                                   | <b>Paper-III</b> | <b>Electron Diffraction:</b> Scattering Intensity versus Scattering Angle   |
|   |                                   | <b>Paper-IV</b>  | Types of fire extinguishers(chemical reaction)  |
| 7 | Vhadade Amar<br>Shripatis.        | <b>Paper-I</b>   | Structures of ABO <sub>3</sub> type, relation between ReO <sub>3</sub> and perovskite BaTiO <sub>3</sub> and its ilmenite structure |
|   |                                   | <b>Paper-II</b>  | Role of metal ions in biological electron transfer processes: iron sulphur proteins   |



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|  |  | <b>Paper-III</b> | <b>Electron Diffraction: Weir Measurement Technique</b> |
|  |  | <b>Paper-IV</b>  | <b>Bioactive materials</b>                              |

**Physical Chemistry**


|   |                                |                  |  |
|---|--------------------------------|------------------|--|
| 1 | Narkar Ankita<br>Sadashiv      | <b>Paper-I</b>   | <b>Introduction:</b> Polymer Science, fundamental terms  |
|   |                                | <b>Paper-II</b>  | Variation of optical and magnetic properties of non material lwith size.   |
|   |                                | <b>Paper-III</b> | The Born–Oppenheimer approximation.  |
|   |                                | <b>Paper-IV</b>  | <b>Thermogravimetry (TG):</b> Principle and Instrumentation  |
| 2 | Shaikh Abu Bakar<br>Bakar      | <b>Paper-I</b>   | Polymer Science historical outline, classification based on: the origin (natural, semi-synthetic, synthetic etc.)            |
|   |                                | <b>Paper-II</b>  | Variation of optical and magnetic properties of non material lwith shape   |
|   |                                | <b>Paper-III</b> | The Born–Oppenheimer approximation   |
|   |                                | <b>Paper-IV</b>  | <b>Thermogravimetry (TG):</b> factors affecting thermo gravimetric curves  |
| 3 | Shaikh Arbaaz Siraj<br>Tamanna | <b>Paper-I</b>   | Polymer Science: the structure (linear, branched, network, hyper branched, dendrimer, ladder, cross linked, IPN),            |
|   |                                | <b>Paper-II</b>  | Variation of optical and magnetic properties of non material with surface characteristics                                    |
|   |                                | <b>Paper-III</b> | LCAO method-molecular orbital formation  |
|   |                                | <b>Paper-IV</b>  | <b>Thermogravimetry (TG):</b> Interpretation of thermo gravimetric curves  |
| 4 | Shaikh Misba<br>Sakhavat Bobby | <b>Paper-I</b>   | The type of atom in the main chain (homo chain, hetero chain)  |
|   |                                | <b>Paper-II</b>  | Variation of optical and magnetic properties of non material with impurities   |
|   |                                | <b>Paper-III</b> | LCAO method-molecular orbital formation  |
|   |                                | <b>Paper-IV</b>  | <b>Thermogravimetry (TG):</b> applications of thermo gravimetry  |
| 5 | Shaikh Mohammed<br>Ammar Ilyas | <b>Paper-I</b>   | The formation (condensation, addition), homo polymers, co polymers (random, alternate, block, graft)                         |
|   |                                | <b>Paper-II</b>  | Relationship between size and shape of nano materials  |
|   |                                | <b>Paper-III</b> | Calculation of energy of hydrogen molecule ion using Valence bond method   |
|   |                                | <b>Paper-IV</b>  | <b>Differential thermal analysis (DTA) and Differential scanning calorimetry (DSC):</b> Principle and instrumentation.       |
| 6 | Shaikh Tubba Ansar             | <b>Paper-I</b>   | the behavior on application of heat (thermoplastic and thermosetting)  |
|   |                                | <b>Paper-II</b>  | Relationship between size and shape of nano materials.   |
|   |                                | <b>Paper-III</b> | Calculation of energy of hydrogen molecule ion using Valence bond method.  |
|   |                                | <b>Paper-IV</b>  | <b>Differential thermal analysis (DTA) and Differential scanning calorimetry (DSC):</b> heat flux and power compensated DSC. |
| 7 | Khan Tabassum Ejaz             | <b>Paper-I</b>   | The form and application (plastics, fibre, elastomers and resins)  |

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|  | Noorjahan | Paper-II  | Less common ions in biology e.g. Ni (urease; structure and reactivity)  |
|  |           | Paper-III | Calculation of energy of hydrogen molecule ion using Valence bond method.   |
|  |           | Paper-IV  | Differential thermal analysis(DTA)and Differential scanning calorimetry (DSC); Interpretation of DTA and DSC curves applications of DTA and DSC |

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M.Sc. Part II- Internal Mark List Sem-III December-2022

**Organic Chemistry**

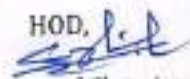
| Sr. No. | Name Of Students             | Paper-I | Paper-II | Paper-III | Paper-IV |
|---------|------------------------------|---------|----------|-----------|----------|
| 1.      | Khan Iqra Azizullah          | 32      | 24       | 28        | 27       |
| 2.      | Khan Khushnuma Abdul         | 30      | 30       | 24        | 32       |
| 3.      | Mansoori Safa Banu Chand     | 20      | 30       | 27        | 29       |
| 4.      | Khan Zainab Parvin Noor      | 30      | 33       | 24        | 32       |
| 5.      | Kumbhar Vaishnavi Shashikant | 28      | 31       | 29        | 25       |
| 6.      | Mishra Nishant Deoprasaksh   | 26      | 30       | 27        | 29       |
| 7.      | Shaikh Zainabfiya Imtiaz     | 27      | 31       | 31        | 31       |

**Inorganic Chemistry**

| Sr. No. | Name Of Students           | Paper-I | Paper-II | Paper-III | Paper-IV |
|---------|----------------------------|---------|----------|-----------|----------|
| 1       | Khan Mehjabeen Mohammad S. | 29      | 30       | 29        | 24       |
| 2       | Mahadeshwar Diksha Dilip   | 30      | 32       | 26        | 25       |
| 3       | Mansuri Md Jahangir N.     | 30      | 32       | 24        | 30       |
| 4       | Neman Kunal Eknath         | 25      | 27       | 24        | 32       |
| 5       | Pandhare Sudhir Bhanudas   | 25      | 27       | 24        | 26       |
| 6       | Singh Shresthi Manoj Kumar | 32      | 31       | 30        | 24       |
| 7       | Vladade Amar Shripatis     | 30      | 29       | 31        | 25       |

**Physical Chemistry**

| Sr. No. | Name Of Students            | Paper-I | Paper-II | Paper-III | Paper-IV |
|---------|-----------------------------|---------|----------|-----------|----------|
| 1       | Khan Tabassum Ejaz N.       | 32      | 31       | 31        | 26       |
| 2       | Narkar Ankita Sadashiv      | 26      | 28       | 27        | 25       |
| 3       | Shaikh Abu Bakar Bakar      | 24      | 30       | 24        | 28       |
| 4       | Shaikh Arbaaz Siraj Tamanna | 26      | 29       | 28        | 24       |
| 5       | Shaikh Misba Sakhat Bobby   | 31      | 27       | 32        | 31       |
| 6       | Shaikh Mohammed Ammar Ilyas | 27      | 26       | 29        | 35       |
| 7       | Shaikh Tubba Ansar          | 33      | 29       | 29        | 30       |

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Department of Chemistry

M.Sc. Part I

Semester II April 2023

| Sr. No | Name of Subject                             | Date and Time                  | Name of Examiner                           |
|--------|---|--------------------------------|--|
| 1      | Physical Chemistry: Course Code: PSCH 201   | 02/04/2023<br>9:00am to 2:00pm | Shri. Yatendra Yadav                       |
| 2      | Inorganic Chemistry: Course Code: PSCH 202  | 03/04/2023<br>9:00am to 2:00pm | Miss. Saira Siddique and<br>Ruqaiya Ansari |
| 3      | Organic Chemistry: Course Code: PSCH 203    | 04/04/2023<br>9:00am to 2:00pm | Dr. Dattatraya Bhangare                    |
| 4      | Analytical Chemistry: Course Code: PSCH 204 | 06/04/2023<br>9:00am to 2:00pm | Shri Pramod Vishwakarma                    |

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Department of Chemistry  
M. Sc. I Sem-II (Internal Presentation Topics) April-2023

| Sr. No. | Name of Students             | Paper No. | Topic For presentation   |
|---------|------------------------------|-----------|--|
| 1.      | Behera Purnima<br>Laxmidhar  | Paper-I   | Maxwell equations, Maxwell thermodynamic Relations; its significance and applications to ideal gases.                    |
|         |                              | Paper-II  | Preparative methods: Chemical methods, Solvothermal,   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: rate and equilibrium constants                                     |
|         |                              | Paper-IV  | Laser as a source of radiation, Fibre optics   |
| 2.      | Gupta Rohit<br>Kumar P.      | Paper-I   | Third law of Thermodynamics, Entropy change for a phase transition, absolute entropies                                   |
|         |                              | Paper-II  | Preparative methods:, Solvothermal, Combustion synthesis,  |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: reaction coordinate diagram, transition state (activated complex), |
|         |                              | Paper-IV  | Introduction of Fourier Transform.   |
| 3.      | Gupta Shivangi<br>Rajendra   | Paper-I   | Joule Thomson experiment, Joule Thomson coefficient  |
|         |                              | Paper-II  | Preparative methods: Microwave, Co-precipitation,  |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: nature of activated complex, Hammond postulate,                    |
|         |                              | Paper-IV  | Derivation of Beer- Lambert's Law and its limitations  |
| 4.      | Khan Hadiye<br>Khatun Abusad | Paper-I   | Joule Thomson coefficient in terms of van der Waals constants  |
|         |                              | Paper-II  | Preparative methods: Langmuir Blodgett (L-B) method,   |
|         |                              | Paper-III | Thermodynamic and kinetic requirements of a reaction: Reactivity vs selectivity,   |
|         |                              | Paper-IV  | factors affecting molecular absorption, types of transitions   |
| 5.      | Khan Parveen<br>Amjad Ali    | Paper-I   | Standard molar entropies and their dependence on molecular mass and molecular structure                                  |
|         |                              | Paper-II  | Biological methods: Synthesis using microorganisms   |
|         |                              | Paper-III | Thermodynamic and kinetic  |

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|     |                            |           |  |
|-----|----------------------------|-----------|--|
|     |                            |           | requirements of a reaction: Curtin-Hammett Principle,  |
|     |                            | Paper-IV  | factors affecting molecular absorption: pH, temperature,   |
| 6.  | Khan Shamina<br>Bano Ansar | Paper-I   | Classical Mechanics, failure of classical mechanics  |
|     |                            | Paper-II  | Applications in the field of semiconductors,   |
|     |                            | Paper-III | Thermodynamic and kinetic requirements of a reaction: Microscopic reversibility,   |
|     |                            | Paper-IV  | Factors affecting molecular absorption: solvent and effect of substituents.  |
| 7.  | Memon Aisha<br>Salim       | Paper-I   | Need for Quantum Mechanics.  |
|     |                            | Paper-II  | Applications in the field of solar cells   |
|     |                            | Paper-III | Thermodynamic and kinetic requirements of a reaction: Kinetic vs thermodynamic control of organic reactions.                                   |
|     |                            | Paper-IV  | Dual spectrometry - Introduction, Principle, Instrumentation and Applications  |
| 8.  | Mishra Ankit<br>Kumar Anil | Paper-I   | Particle waves and Schrödinger wave equation   |
|     |                            | Paper-II  | Characterisation of Coordination compounds: Formation, thermal studies, Conductivity measurement   |
|     |                            | Paper-III | Ester hydrolysis: Classification   |
|     |                            | Paper-IV  | Applications of Ultraviolet and Visible spectroscopy: 1) On charge transfer absorption   |
| 9.  | Mishra Shishir<br>Santosh  | Paper-I   | wave functions, properties of wave functions   |
|     |                            | Paper-II  | Characterisation of Coordination compounds: electronic spectral and magnetic measurements  |
|     |                            | Paper-III | Determining mechanism of a reaction: Product analysis,   |
|     |                            | Paper-IV  | Applications of Ultraviolet and Visible spectroscopy: Simultaneous spectroscopy  |
| 10. | Mishra Utkarsh<br>Narendra | Paper-I   | Normalization of wave functions,   |
|     |                            | Paper-II  | Characterisation of Coordination compounds: IR   |
|     |                            | Paper-III | Determining mechanism of a reaction: kinetic studies, use of isotopes (Kinetic isotope effect - primary and secondary kinetic isotope effect). |
|     |                            | Paper-IV  | Applications of Ultraviolet and  |



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|-----|-----------------------------------|-----------|---|
|     |                                   |           | Visible spectroscopy: Derivative Spectroscopy   |
| 11. | Khan Roshan<br>Jahan<br>Kyamuddin | Paper-I   | Orthogonality of wave functions.  |
|     |                                   | Paper-II  | Characterisation of Coordination compounds: NMR   |
|     |                                   | Paper-III | Determining mechanism of a reaction: Detection and trapping of intermediates,   |
|     |                                   | Paper-IV  | Oxidation number, rules for assigning oxidation number,   |
| 12. | Shaikh Afrin<br>Mushtaq           | Paper-I   | Operators and their algebra   |
|     |                                   | Paper-II  | Preparative methods: Chemical methods   |
|     |                                   | Paper-III | Characterisation of Coordination compounds: ESR   |
|     |                                   | Paper-IV  | Redox reaction in term of oxidation number, oxidizing and reducing agents,  |
| 13. | Shaikh Ayesha<br>Riyaz            | Paper-I   | Linear and Hermitian operators  |
|     |                                   | Paper-II  | Microwave synthesis (discussion on principles, examples, merits and demerits are expected)                                    |
|     |                                   | Paper-III | Determining mechanism of a reaction: crossover experiments and stereochemical evidence.                                       |
|     |                                   | Paper-IV  | Equivalent weight of oxidizing and reducing agents, stoichiometry of redox titration  |
| 14. | Shaikh Salim<br>Dilbahar          | Paper-I   | Operators for the dynamic variables of a system such as, position, linear momentum, angular momentum                          |
|     |                                   | Paper-II  | Methods of preparation for inorganic solids: Ceramic method, precursor method, sol-gel method (applications in Biosensors),   |
|     |                                   | Paper-III | Determining mechanism of a reaction: stereochemical evidence.   |
|     |                                   | Paper-IV  | Normality of a solution of a oxidizing / reducing agent and its relationship with molarity.                                   |
| 15. | Shaikh<br>Mohammad<br>Sufiyan M.  | Paper-I   | Operators for the dynamic variables of a system such as, total energy, eigen functions, eigen values and eigen value equation |
|     |                                   | Paper-II  | Layer structure [cadmium chloride and iodide ( $\text{CdCl}_2$ , $\text{CdI}_2$ )].   |
|     |                                   | Paper-III | Acids and Bases: Factors affecting acidity and basicity: Electronegativity  |
|     |                                   | Paper-IV  | Calculations of pH of acids, bases, acidic and basic buffers.   |
| 16. | Shingada                          | Paper-I   | Schrödinger wave equation as the  |

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|     |                           |           |   |
|-----|---------------------------|-----------|---|
|     | Chandresh Radaka          |           | eigen value equation of the Hamiltonian operator  |
|     |                           | Paper-II  | Electronic structure of solids and band theory  |
|     |                           | Paper-III | Acids and Bases: Factors affecting acidity and basicity: inductive effect   |
|     |                           | Paper-IV  | Calculations of pH of acids, bases, acidic and basic buffers.   |
| 17. | Siddiqui Hajra Dilshad A. | Paper-I   | Average value and the expectation value of a dynamic variable of the system   |
|     |                           | Paper-II  | Fermi level, K Space and Brillouin Zones.   |
|     |                           | Paper-III | Acids and Bases: Factors affecting acidity and basicity: resonance, bond strength, electrostatic effects,                 |
|     |                           | Paper-IV  | Solubility and solubility equilibria, effect of presence of common ion.   |
| 18. | Singh Vinay Ramesh        | Paper-I   | Postulates of Quantum Mechanics   |
|     |                           | Paper-II  | Structures of Compounds of the type: AB [nickel arsenide (NiAs)], AB <sub>2</sub> [fluorite (CaF <sub>2</sub> )]          |
|     |                           | Paper-III | Acids and Bases: Factors affecting acidity and basicity: hybridization, aromaticity and solvation.                        |
|     |                           | Paper-IV  | Stoichiometry of chemical reactions, concept of kg mol, limiting reactant, theoretical and practical yield.               |
| 19. | Tiwari Anjali Rajesh      | Paper-I   | Schrodinger's Time independent wave equation  |
|     |                           | Paper-II  | Structures of Compounds of the type: AB [nickel arsenide (NiAs)],)  |
|     |                           | Paper-III | Comparative study of acidity and basicity of organic compounds on the basis of pK <sub>a</sub> values.                    |
|     |                           | Paper-IV  | Calculations of ppm, ppb and dilution of the solutions, concept of mmol   |
| 20. | Yadav Mansi Keshav        | Paper-I   | Schrodinger's time dependent wave equation.   |
|     |                           | Paper-II  | Structures of Compounds of the type: AB <sub>2</sub> [fluorite (CaF <sub>2</sub> )]                                       |
|     |                           | Paper-III | Leveling effect and non-aqueous solvents.   |
|     |                           | Paper-IV  | Nomenclature of all eight mechanisms of acid and base catalyzed hydrolysis.   |
| 21. | Yadav Neha Pramod         | Paper-I   | Application of quantum mechanics to the following systems: a) Free particle, wave function and energy of a free particle. |



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|  |  | Paper-II  | Structures of Compounds of the type:<br>AB <sub>2</sub> rutile (TiO <sub>2</sub> ) structure |
|  |  | Paper-III | Acid and base catalysis - general and<br>specific catalysis with examples.                   |
|  |  | Paper-IV  | Concentration of a solution based on<br>volume and mass units.                               |

  
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Department of Chemistry  
M. Sc. I Sem I (Internal Marks) April-2023

| Sr. No. | Name of Students              | Paper-I | Paper-II | Paper-III | Paper-IV |
|---------|-------------------------------|---------|----------|-----------|----------|
| 1.      | KHAN IQRA AZIZULLAHA          | 29      | 26       | 28        | 30       |
| 2.      | KHAN KHUSHNUMA ABDUL          | 32      | 26       | 31        | 30       |
| 3.      | KHAN MEHJABEEN MOHAMMAD SHAFI | 30      | 24       | 22        | 31       |
| 4.      | KHAN TABASSUM EJAZ            | 31      | 25       | 23        | 29       |
| 5.      | KHAN ZAINAB PARVIN NOOR       | 33      | 26       | 26        | 30       |
| 6.      | KUMBHAR VAISHNAVI SHASHIKANT  | 32      | 27       | 27        | 31       |
| 7.      | MAHADESHWAR DIKSHA DILIP      | 27      | 26       | 24        | 32       |
| 8.      | MANSOORI SAFA BANU CHAND      | 32      | 26       | 26        | 30       |
| 9.      | ANSURI MD JAHANGIR NIZAMUDDIN | 30      | 26       | 25        | 29       |
| 10.     | MISHRA NISHANT DEOPRAKASH     | 35      | 30       | 29        | 32       |
| 11.     | NARKAR ANKITA SADASHIV        | 27      | 24       | 27        | 28       |
| 12.     | NEMAN KUNAL EKNATH            | 32      | 23       | 25        | 30       |
| 13.     | PANDHARE SUDHIR BHANUDAS      | 30      | 24       | 23        | 29       |
| 14.     | SHAIKH ABU BAKAR BAKAR HUSAIN | 31      | 25       | 22        | 30       |
| 15.     | SHAIKH ARBAAB SIRAJ           | 27      | 22       | 27        | 30       |
| 16.     | SHAIKH MISRA SAKHAVAT         | 30      | 26       | 28        | 31       |
| 17.     | SHAIKH MOHAMMED AMMAR ILYAS   | 31      | 26       | 25        | 31       |
| 18.     | SHAIKH TUBBA ANSAR            | 30      | 25       | 24        | 30       |
| 19.     | SHAIKH ZAINABALFIYA IMTIAZ    | 35      | 32       | 27        | 31       |
| 20.     | SINGH SHRASTI MANOJ KUMAR     | 30      | 30       | 30        | 32       |
| 21.     | VRADADE AMAR SHRIPATI         | 26      | 26       | 23        | 30       |

  
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Fax (Principal) 022-28202188


Department of Chemistry

M.Sc. Part II

Semester IV April 2023

Physical Chemistry

| Sr. No | Subject Code | Name of Subject   | Date and Time                   | Name of Examiner     |
|--------|--------------|---|---------------------------------|----------------------|
| 1      | PSCHP401     | Chemistry: Polymer, Green, Biophysical and Applied.       | 22/04/2023<br>9:00am to 11:00am | Shri. Yatendra Yadav |
| 2      | PSCHP402     | Material Science, Network And Irreversible Thermodynamics | 22/04/2023<br>12:00pm to 2:00pm | Shri. Rajeev Verma   |
| 3      | PSCHP403     | Symmetry & Spectroscopy                                   | 24/04/2023<br>9:00am to 11:00am | Miss. Azra Ansari    |
| 4      | PSCHP404     | Research Methodology                                      | 24/04/2023<br>12:00pm to 2:00pm | Shri. Akash Maurya   |

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Department of Chemistry

M.Sc. Part II

Semester IV April 2023

Inorganic Chemistry

| Sr. No | Subject Code | Name of Subject                                  | Date and Time                   | Name of Examiner     |
|--------|--------------|--|---------------------------------|----------------------|
| 1      | PSCHP401     | Properties of Inorganic Solids and Group Theory. | 25/04/2023<br>9:00am to 11:00am | Miss. Saira Siddique |
| 2      | PSCHP402     | Organometallics and main group Chemistry         | 25/04/2023<br>12:00pm to 2:00pm | Miss. Ruqaiya Ansari |
| 3      | PSCHP403     | Instrumental methods in Inorganic Chemistry      | 26/04/2023<br>9:00am to 11:00am | Miss. Priyanka Sonar |
| 4      | PSCHP404     | Research Methodology                             | 26/04/2023<br>12:00pm to 2:00pm | Shri. Rohan Narkar   |

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**CHEMISTRY**





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**ISMAIL YUSUF COLLEGE OF ARTS, SCIENCE AND COMMERCE**  
**NAAC 'A' Grade**  
**Jogeshwari (East), Mumbai 400 060**

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Department of Chemistry

M.Sc. Part II

Semester IV April 2023

Organic Chemistry

| Sr. No | Subject Code | Name of Subject                             | Date and Time                   | Name of Examiner         |
|--------|--------------|---|---------------------------------|--------------------------|
| 1      | PSCHO401     | Theoretical organic chemistry-II            | 28/04/2023<br>9:00am to 11:00am | Miss. Pratiksha Borase   |
| 2      | PSCHO402     | Synthetic organic chemistry-II              | 28/04/2023<br>12:00pm to 2:00pm | Dr. Kiran Taksande       |
| 3      | PSCHO403     | Natural products and heterocyclic chemistry | 29/04/2023<br>9:00am to 11:00am | Dr. Dattatraya Bhangare  |
| 4      | PSCHO404     | Research Methodology                        | 29/04/2023<br>12:00pm to 2:00pm | Shri. Pramod Vishwakarma |

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**Presentation Topics for M.Sc. II Sem-IV Internal April-2023**

Organic Chemistry

| Sr. No. | Name of Students               | Paper No. | Topic For presentation   |
|---------|--------------------------------|-----------|--|
| 1.      | Harwani Seleni Sunil           | Paper-I   | Structural effects and reactivity: Linear free energy relationship (LFER) in determination of organic reaction mechanism.          |
|         |                                | Paper-II  | <b>Protecting groups in Organic Synthesis:</b> Protection and deprotection of the hydroxyl.  |
|         |                                | Paper-III | <b>Steroids:</b> General structure, classification   |
|         |                                | Paper-IV  | History and evolution of cheminformatics   |
| 2.      | Khan Nikhat Parveen Ahsanullah | Paper-I   | The Hammett equation, substituent constants  |
|         |                                | Paper-II  | <b>Protecting groups in Organic Synthesis:</b> Protection and deprotection of the carbonyl   |
|         |                                | Paper-III | <b>Steroids:</b> Occurrence, biological role, important structural and stereochemical features of: corticosteroids.                |
|         |                                | Paper-IV  | Use of Cheminformatics.  |
| 3.      | Khan Umar Farooq Khan          | Paper-I   | Theories of substituent effects, interpretation of $\sigma$ -values  |
|         |                                | Paper-II  | <b>Protecting groups in Organic Synthesis:</b> Protection and deprotection of the amino.   |
|         |                                | Paper-III | <b>Steroids:</b> Occurrence, biological role, important structural and stereochemical features of: steroidal hormones              |
|         |                                | Paper-IV  | Prospects of cheminformatics   |
| 4.      | Sidique Mohamad Ahmad Abdul    | Paper-I   | Theories of substituent effects, interpretation of $\sigma$ -values  |
|         |                                | Paper-II  | <b>Protecting groups in Organic Synthesis:</b> Protection and deprotection of the carboxyl functional groups and its applications. |
|         |                                | Paper-III | <b>Steroids:</b> Occurrence, biological role, important structural and stereochemical features of: steroidal                       |



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|    |                                 |           |  |
|----|---------------------------------|-----------|--|
|    |                                 |           | alkaloids.   |
|    |                                 | Paper-IV  | Molecular modeling and structure elucidation.  |
| 5. | Tiwari Sarvagya<br>Kumar Arvind | Paper-I   | Uses of Hammett equation, deviations from Hammett equation.  |
|    |                                 | Paper-II  | <b>Concept of umpolung (Reversal of polarity):</b> Generation of acyl anion equivalent using 1,3-dithianes, methyl thiomethyl sulfoxides |
|    |                                 | Paper-III | <b>Steroids:</b> Occurrence, biological role, important structural and stereochemical features of: sterols and bile acids.               |
|    |                                 | Paper-IV  | <b>Representation of molecules and chemical reactions:</b> Nomenclature, Different types of notations                                    |
| 6. | Yadav Anish<br>Anil             | Paper-I   | Neighbouring group participation: Mechanism and effects of anchimeric assistance.  |
|    |                                 | Paper-II  | Multicomponent reactions: Strecker Synthesis, Ugi 4CC  |
|    |                                 | Paper-III | General structural features, occurrence, biological importance and applications of: carotenoids  |
|    |                                 | Paper-IV  | Shikimic Acid pathway: Biosynthesis of shikimic acid   |
| 7. | Yadav Dinesh<br>Suresh          | Paper-I   | NGP by unshared/ lone pair electrons, n-electrons, aromatic rings  |
|    |                                 | Paper-II  | Multicomponent reactions: Biginelli synthesis, Hantzsch synthesis  |
|    |                                 | Paper-III | General structural features, occurrence, biological importance and applications of: anthocyanins   |
|    |                                 | Paper-IV  | Shikimic Acid pathway: Biosynthesis of aromatic amino acids, cinnamic acid and its derivatives   |

**Inorganic Chemistry**

|   |                       |           |  |
|---|-----------------------|-----------|--|
| 1 | Beig Parvej<br>KhwaJa | Paper-I   | <b>Electrical properties of solids:</b> Conductivity: Solid Electrolytes; Fast Ion |
|   |                       | Paper-II  | <b>Organometallic Chemistry:</b> Metal-Metal Bonding and Metal Clusters            |
|   |                       | Paper-III | <b>Infrared spectroscopy:</b> Fundamental modes of vibrations, selection rules     |
|   |                       | Paper-IV  | Primary, Secondary and Tertiary sources  |

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|---|---------------------------|------------------|--|
| 2 | Kadam Pratik<br>Prakash   | <b>Paper-I</b>   | Conductors; Mechanism of Conductivity; Hopping Conduction.   |
|   |                           | <b>Paper-II</b>  | <b>Organometallic Chemistry:</b> Isolobal Analogy  |
|   |                           | <b>Paper-III</b> | <b>Infrared spectroscopy:</b> IR absorption bands of metal - donor atom  |
|   |                           | <b>Paper-IV</b>  | Journal abbreviations, abstracts, current titles, reviews  |
| 3 | Mane Swapnali<br>Yuvraj   | <b>Paper-I</b>   | <b>Other Electrical Properties:</b> Thomson and Seebeck Effects;   |
|   |                           | <b>Paper-II</b>  | <b>Organometallic Chemistry:</b> Organo Palladium Complexes (preparations, properties and applications.)                       |
|   |                           | <b>Paper-III</b> | <b>Infrared spectroscopy:</b> effect of complexation on the IR spectrum of ligands formations on the IR of ligands like $NH_3$ |
|   |                           | <b>Paper-IV</b>  | monographs, dictionaries, textbooks, current contents  |
| 4 | Gureshi Heena<br>Jesar    | <b>Paper-I</b>   | Thermocouples and their Applications;  |
|   |                           | <b>Paper-II</b>  | <b>Organometallic Chemistry:</b> Organo. Platinum Complexes (preparations, properties and applications)                        |
|   |                           | <b>Paper-III</b> | <b>Infrared spectroscopy:</b> effect of complexation on the IR spectrum of ligands formations on the IR of ligands like $CN^-$ |
|   |                           | <b>Paper-IV</b>  | Introduction to Chemical Abstracts and Beilstein, Subject Index other Indices with examples.                                   |
| 5 | Singh Ashwini<br>Anil     | <b>Paper-I</b>   | Hall Effect; Dielectric, Ferroelectric,  |
|   |                           | <b>Paper-II</b>  | Substance Index, Author Index, Formula Index   |
|   |                           | <b>Paper-III</b> | <b>Infrared spectroscopy:</b> effect of complexation on the IR spectrum of ligands formations on the IR of ligands like CO     |
|   |                           | <b>Paper-IV</b>  | Web sources, E-journals, Journal access, TOC alerts  |
| 6 | Khan Mohd<br>Shaban Abdul | <b>Paper-I</b>   | Hall Effect; Piezoelectric and Pyroelectric  |
|   |                           | <b>Paper-II</b>  | <b>Organometallic Chemistry:</b> (a) Metal-Metal Bonding and Metal Clusters  |
|   |                           | <b>Paper-</b>    | <b>Raman spectroscopy:</b> Raman   |



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|---|--------------------------|-----------|--|
| 7 | Manihar<br>Rafiqullah S. | III       | spectroscopy for diatomic molecules  |
|   |                          | Paper-IV  | Hot articles, Citation Index, Impact factor, H-index, E-consortium           |
|   |                          | Paper-I   | Materials and their Inter-relationships                                      |
|   |                          | Paper-II  | Catalysis-Homogenous and Heterogenous Catalysis: Comparison.                 |
|   |                          | Paper-III | Determination of molecular structures like diatomic and triatomic molecules. |
|   |                          | Paper-IV  | Bioactive materials  |

Physical Chemistry 2022-23

|   |                                      |           |  |
|---|--------------------------------------|-----------|--|
| 1 | Narkar<br>Ankita<br>Sadashiv         | Paper-I   | <b>Polymers in solid state</b> - Transitions (glass transition and crystalline melting temperature)                          |
|   |                                      | Paper-II  | <b>Metals and alloys:</b> Solidification of metals and alloys-homogeneous and heterogeneous nucleation<br>Growth of crystals |
|   |                                      | Paper-III | Recapitulation: point groups, character tables.  |
|   |                                      | Paper-IV  | weighted linear case, analysis of residuals  |
| 2 | Shaikh Abu<br>Bakar Bakar            | Paper-I   | <b>Polymers in solid state</b> crystalline behaviour, factors affecting crystallinity  |
|   |                                      | Paper-II  | <b>Metals and alloys:</b> Variation of optical and magnetic properties of non material with shape.                           |
|   |                                      | Paper-III | Recapitulation: character tables   |
|   |                                      | Paper-IV  | fitting of linear equations, simple linear cases   |
| 3 | Shaikh<br>Arbaaz<br>Siraj<br>Tamanna | Paper-I   | <b>Polymers in solid state:</b> polymer blends and Alloys.   |
|   |                                      | Paper-II  | <b>Metals and alloys:</b> growth of silicon single crystal   |
|   |                                      | Paper-III | Reduction formula, application of reduction formula to vibrational modes of water molecule                                   |
|   |                                      | Paper-IV  | Correlation and regression, curve fitting  |
| 4 | Shaikh<br>Misba<br>Sakhavat<br>Bobby | Paper-I   | <b>Identification and characterization of polymers:</b> Chemical analysis- End group analysis                                |
|   |                                      | Paper-II  | <b>Metals and alloys:</b> growth of silicon single crystal   |
|   |                                      | Paper-III | Reduction formula, application of  |

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|   |                                       |           |  |
|---|---------------------------------------|-----------|--|
|   |                                       |           | reduction formula to vibrational modes of water molecule   |
|   |                                       | Paper-IV  | Chemometrics, Analysis of Variance (ANOVA)   |
| 5 | Shaikh<br>Mohammed<br>Ammar Ilyas     | Paper-I   | Physical analysis by Spectral methods: IR, UV.   |
|   |                                       | Paper-II  | Metallic solid solutions-substitutional and interstitial solid solutions   |
|   |                                       | Paper-III | Application in vibrational spectroscopy, selection rules for IR spectroscopy for molecules such as H <sub>2</sub> O. |
|   |                                       | Paper-IV  | Descriptive statistics, choosing and using statistical tests   |
| 6 | Shaikh<br>Tubba Ansar                 | Paper-I   | Physical analysis by Spectral methods:, Ramam, NMR, X-ray Diffraction analysis                                       |
|   |                                       | Paper-II  | Metallic solid solutions-substitutional and interstitial solid solutions.  |
|   |                                       | Paper-III | Application in vibrational spectroscopy, selection rules for IR spectroscopy for molecules such as CO <sub>2</sub>   |
|   |                                       | Paper-IV  | Scientific methods and design of experiments.  |
| 7 | Khan<br>Tabassum<br>Ejaz<br>Noorjahan | Paper-I   | Physical analysis by Spectral methods: Microscopic methods: SEM, TEM,  |
|   |                                       | Paper-II  | Crystalline imperfections-point.   |
|   |                                       | Paper-III | Application in vibrational spectroscopy, selection rules for IR spectroscopy for molecules such as HF                |
|   |                                       | Paper-IV  | SI units and their use   |

  
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**M.Sc. Part II-(Sem-IV) Internal Mark List April-2023**

**Organic Chemistry**


| Sr. No. | Name of Students             | Paper-I | Paper-II | Paper-III | Paper-IV |
|---------|------------------------------|---------|----------|-----------|----------|
| 1.      | KHAN IQRA AZIZULLAHA         | 32      | 24       | 28        | 27       |
| 2.      | KHAN KHUSHNUMA ABDUL         | 30      | 30       | 24        | 32       |
| 3.      | MANSOORI SAFA BANU CHAND     | 20      | 30       | 27        | 29       |
| 4.      | KHAN ZAINAB PARVIN NOOR      | 30      | 33       | 24        | 32       |
| 5.      | KUMBHAR VAISHNAVI SHASHIKANT | 28      | 31       | 29        | 25       |
| 6.      | MISHRA NISHANT DEOPRAKASH    | 26      | 30       | 27        | 29       |
| 7.      | SHAIKH ZAINABALFIYA IMTIAZ   | 27      | 31       | 31        | 31       |

**Inorganic Chemistry**

| Sr. No. | Name of Students         | Paper-I | Paper-II | Paper-III | Paper-IV |
|---------|--------------------------|---------|----------|-----------|----------|
| 1       | KHAN MENJABEEN MOHAMMAD  | 29      | 30       | 29        | 24       |
| 2       | MAHADESHWAR DIKSHA DILIP | 30      | 32       | 26        | 25       |
| 3       | MANSURI MD JAHANGIR N.   | 30      | 32       | 24        | 30       |
| 4       | NEMAN KUNAL EKNATH       | 25      | 27       | 24        | 32       |
| 5       | PANDHARE SUDHIR BHANUDAS | 25      | 27       | 24        | 26       |
| 6       | SINGH SHRISTI MANOJ      | 32      | 31       | 30        | 24       |
| 7       | VHADADE AMAR SHRIPATIS   | 30      | 29       | 31        | 25       |

**Physical Chemistry**

| Sr. No. | Name of Students         | Paper-I | Paper-II | Paper-III | Paper-IV |
|---------|--------------------------|---------|----------|-----------|----------|
| 1       | KHAN TABASSUM EJAZ N.    | 32      | 31       | 31        | 26       |
| 2       | NARKAR ANKITA SADASHIV   | 26      | 28       | 27        | 25       |
| 3       | SHAIKH ABU BAKAR BAKAR   | 24      | 30       | 24        | 28       |
| 4       | SHAIKH ARBAAZ SIRAJ T.   | 26      | 29       | 28        | 24       |
| 5       | SHAIKH MISBA SAKHAVAT B. | 31      | 27       | 32        | 31       |
| 6       | SHAIKH MOHAMMED AMMAR I. | 27      | 26       | 29        | 35       |
| 7       | SHAIKH TUBBA ANSAR       | 33      | 29       | 29        | 30       |

  
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